



# 流动化学在光催化领域的应用

报告人：潘辉

导师：左智伟 研究员

中国科学院上海有机化学研究所

2023年5月13日

# 提纲



一

可见光催化的重要性和局限性

二

流动化学的基础知识

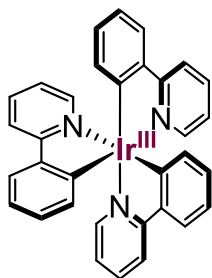
三

常见的泵和连续流光反应器

四

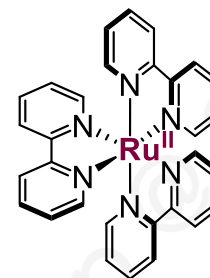
连续流条件下铈-光催化反应的放大应用研究

## 2.1 研究背景：光催化反应的重要性



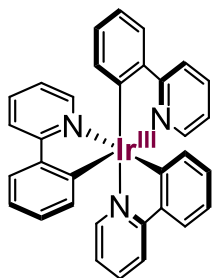
$\text{Ir}^{\text{III}}(\text{ppy})_3$

- Chemically stable
- Visible light absorption
- Long-lived MLCT excited state
- Effective oxidant and reductant



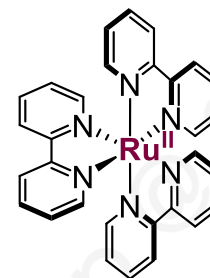
$\text{Ru}^{\text{II}}(\text{bpy})_3^{2+}$

# 1.1 光催化反应的重要性



$\text{Ir}^{\text{III}}(\text{ppy})_3$

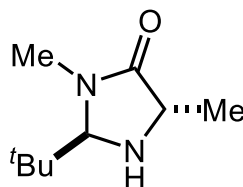
- Chemically stable
- Visible light absorption
- Long-lived MLCT excited state
- Effective oxidant and reductant



$\text{Ru}^{\text{II}}(\text{bpy})_3^{2+}$



Transition-metal  
catalysis



Enantioselective  
catalysis



Electrocatalysis

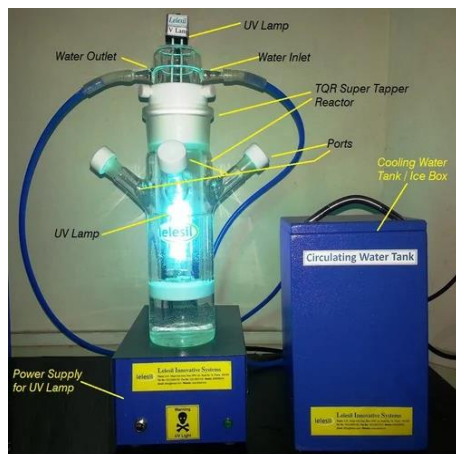


Biocatalysis

D. W. C. MacMillan, et al. *J. Org. Chem.* **2016**, *81*, 6898–6926.  
T. Noël, et al. *Chem. Rev.* **2022**, *122*, 2752–2906.

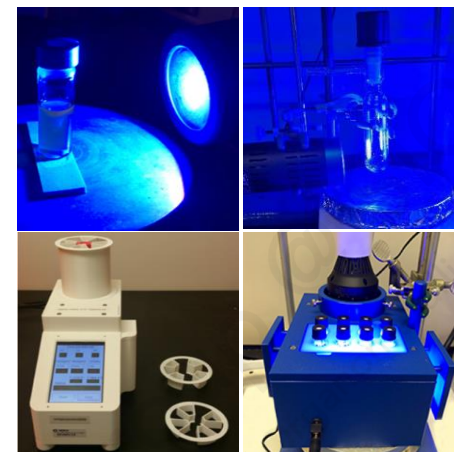
## 2.1 研究背景：光催化反应的局限性

### Traditional UV Photoreaction

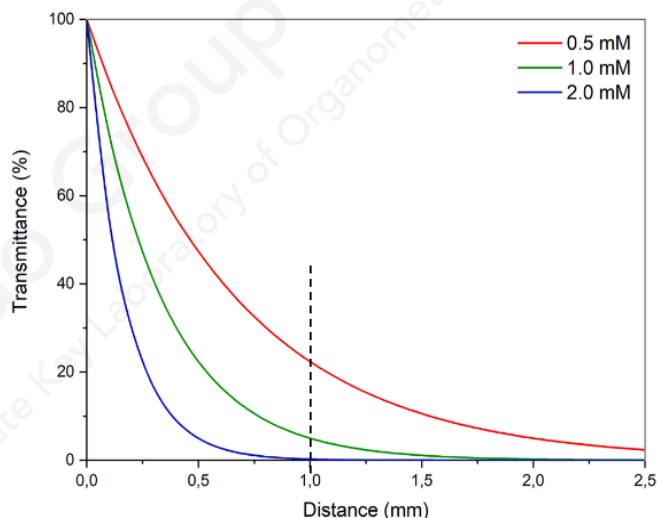


- Low selectivity
- High cost light sources
- Scale-up issues
- Highly selective and powerful
- Energy efficient and durable LED

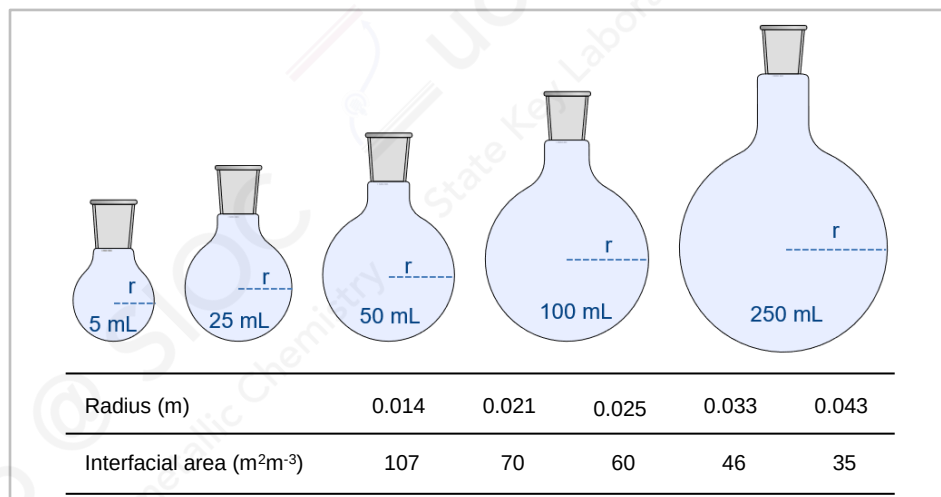
### Visible-Light Photocatalysis



### Scale-up issues



**Figure 4.** Attenuation of light as a function of distance in a photocatalytic reaction using  $\text{Ru}(\text{bpy})_3\text{Cl}_2$  ( $c = 0.5, 1, \text{ and } 2 \text{ mM}$ ,  $\epsilon = 13\,000 \text{ cm}^{-1}\cdot\text{M}^{-1}$ ) utilizing the Bouguer–Lambert–Beer correlation

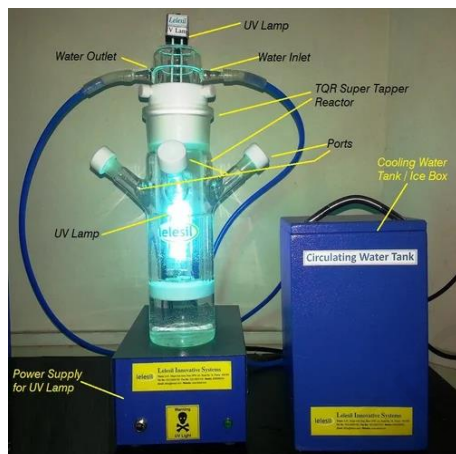


R. Baxendale, et al. *Org. Process Res. Dev.* **2016**, 20, 327–360.

T. Noël, et al. *Chem. Rev.* **2022**, 122, 2752–2906.

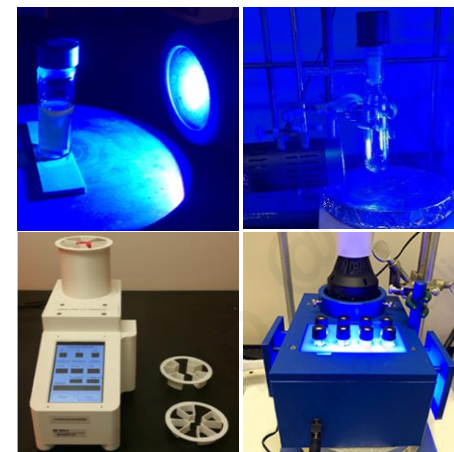
## 2.1 研究背景：光催化反应的局限性

### Traditional UV Photoreaction

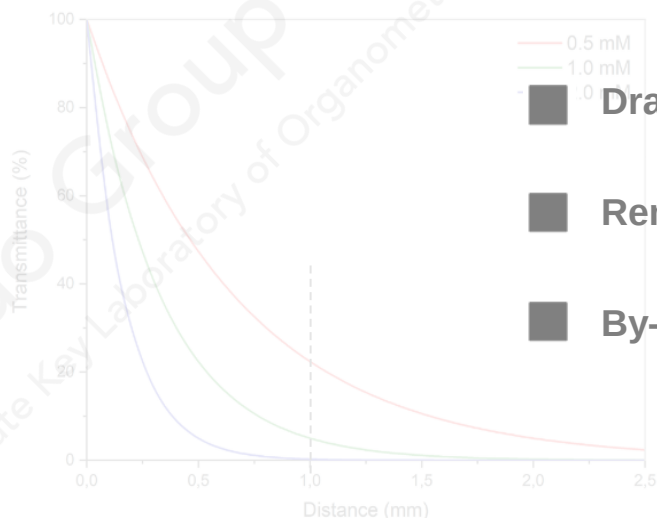


- Low selectivity
- High cost light sources
- Scale-up issues
- Highly selective and powerful
- Energy efficient and durable LED

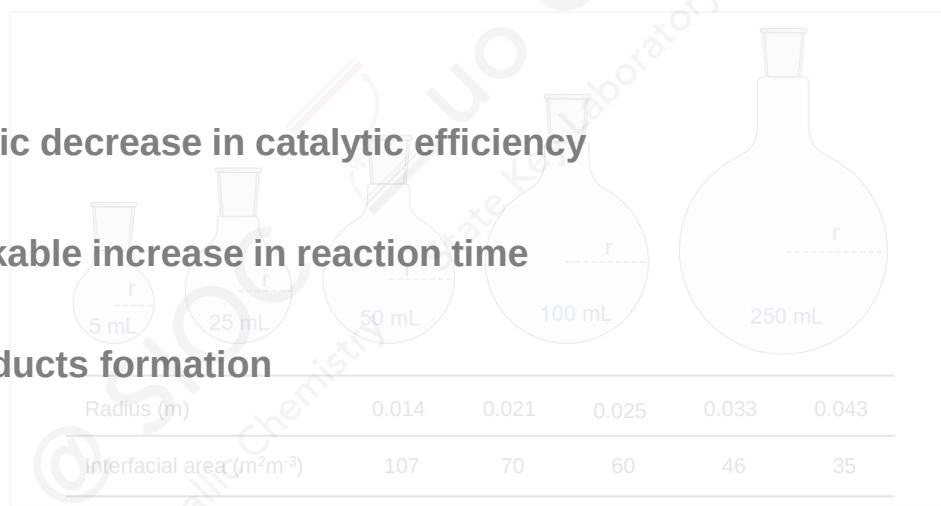
### Visible-Light Photocatalysis



### Scale-up issues



- Dramatic decrease in catalytic efficiency
- Remarkable increase in reaction time
- By-products formation



**Figure 4.** Attenuation of light as a function of distance in a photocatalytic reaction using  $\text{Ru}(\text{bpy})_3\text{Cl}_2$  ( $c = 0.5, 1, \text{ and } 2 \text{ mM}$ ,  $\epsilon = 13\,000 \text{ cm}^{-1}\cdot\text{M}^{-1}$ ) utilizing the Bouguer–Lambert–Beer correlation

R. Baxendale, et al. *Org. Process Res. Dev.* **2016**, *20*, 327–360.

T. Noël, et al. *Chem. Rev.* **2022**, *122*, 2752–2906.

# 提纲

一

可见光催化的重要性和局限性



二

流动化学的基础知识

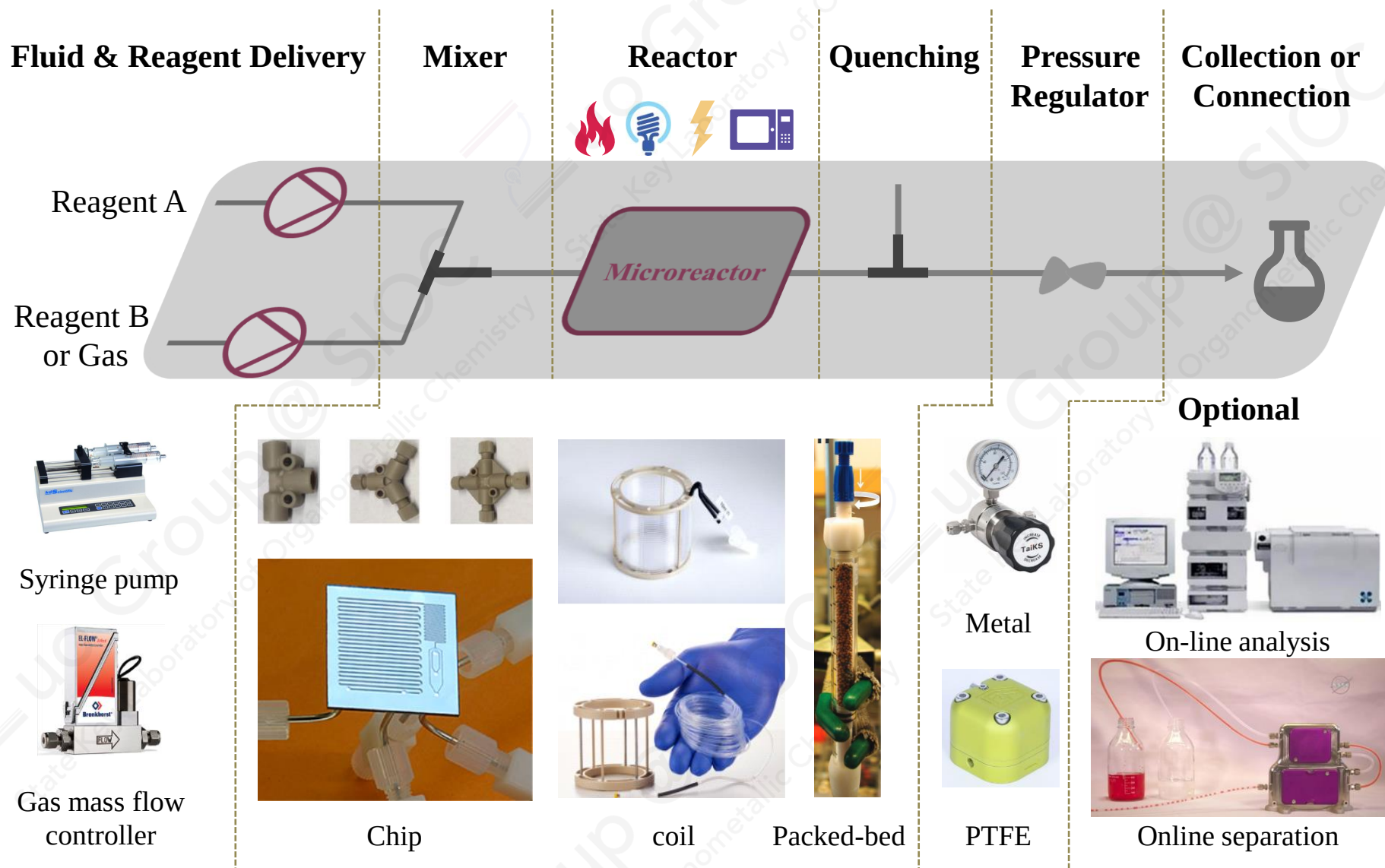
三

常见的泵和连续流光反应器

四

连续流条件下铈-光催化反应的放大应用研究

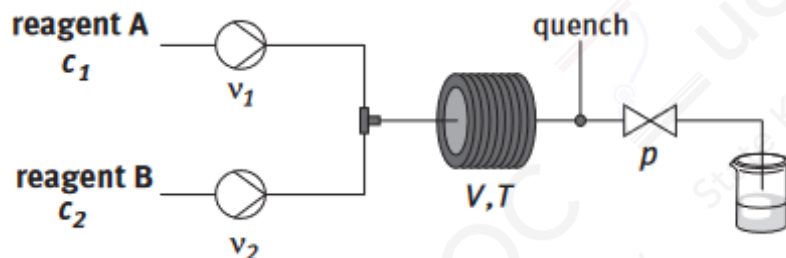
## 2.1 流动化学装置





## 2.2 流动反应与间歇式反应的区别

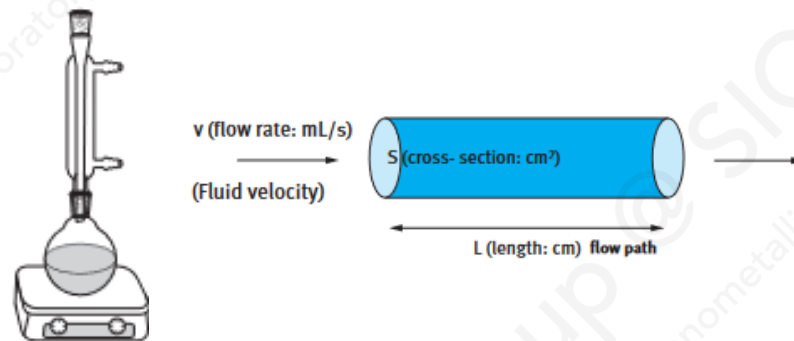
### Stoichiometry



Batch:  $c, \eta$

Flow:  $c, \eta, \mathbf{v}$

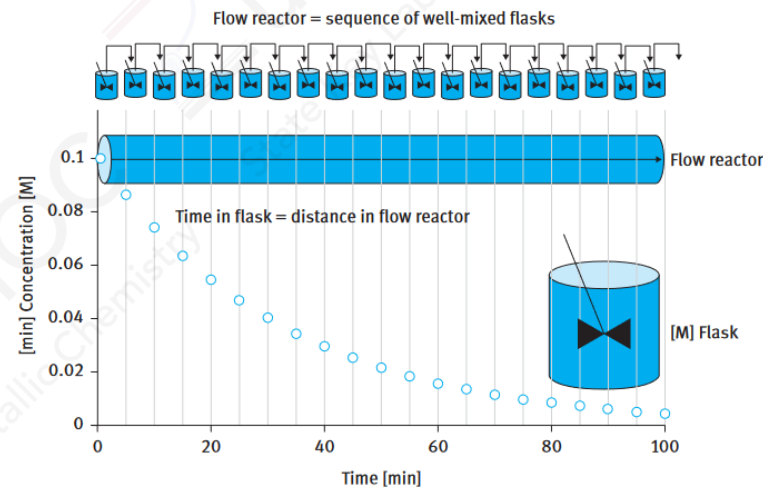
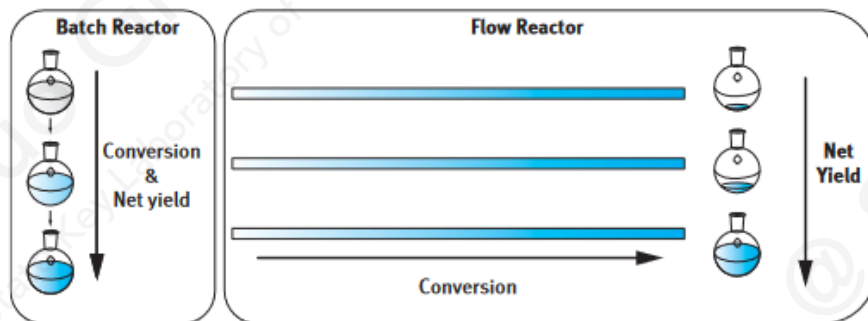
### Reaction time & Residence time



Batch:  $t$

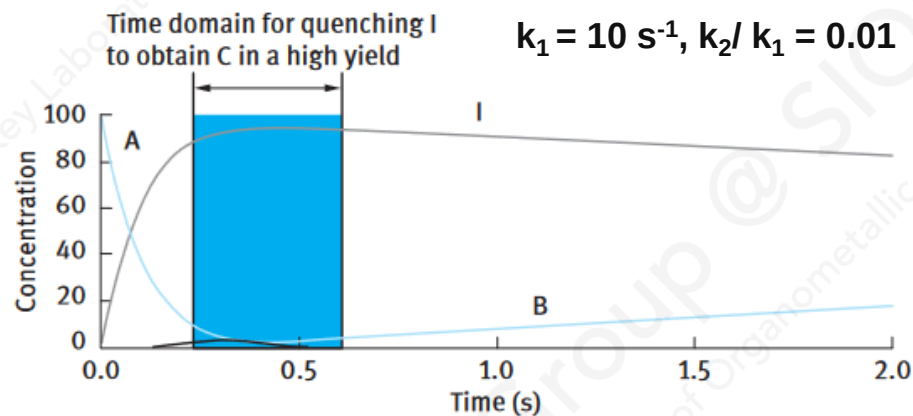
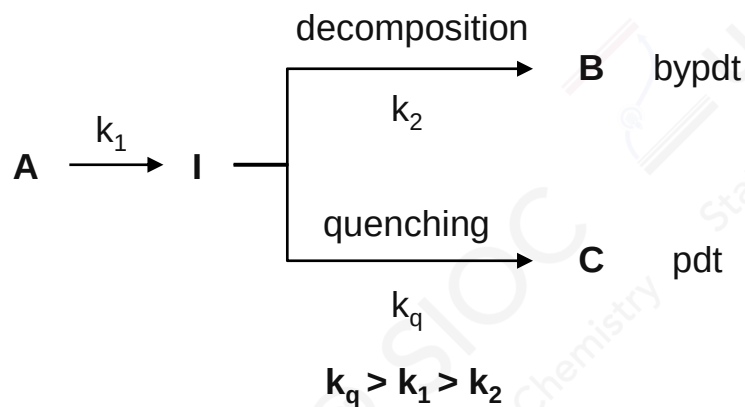
$$\text{Flow: } t_R = \frac{V}{v_1 + v_2} \quad t_R = \frac{V}{v_L + v_G/P}$$

### Reaction progress & Steady-state characteristics



## 2.3 流动化学的特征和优势

### Continuous-flow characteristics



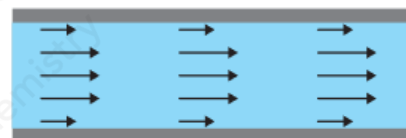
### Precise residence time control

### Large specific surface area ( $d < 1 \text{ mm}$ , large $S$ )



|                  |           |
|------------------|-----------|
| Size             | 1/100     |
| Surface area     | 1/10000   |
| Volume           | 1/1000000 |
| Surface / Volume | 100       |

Laminar flow



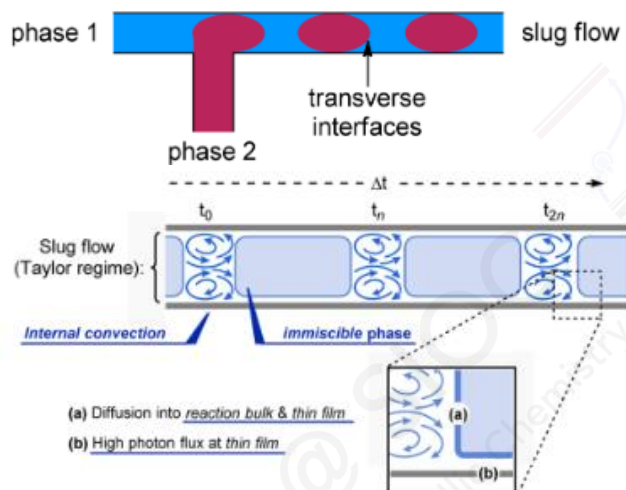
Diffusion time

$$t_d = \frac{d^2}{D}$$

### Uniform mixing: flow (ms to s) vs. batch (s to h)

## 2.3 流动化学的特征和优势

### Large specific surface area ( $d < 1 \text{ mm}$ , large $S$ )

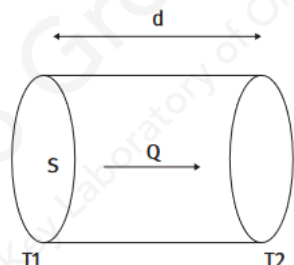


**Table 1. Interfacial Surface Areas for Various Reactors<sup>a</sup>**

| type of reactor                            | interfacial area ( $\text{m}^2 \text{m}^{-3}$ ) |
|--------------------------------------------|-------------------------------------------------|
| 5 mL round-bottom flask (rbf) <sup>b</sup> | 141                                             |
| 50 mL rbf <sup>b</sup>                     | 66                                              |
| 250 mL rbf <sup>b</sup>                    | 38                                              |
| tube reactors, horizontal and coiled       | 50–700                                          |
| tube reactors, vertical                    | 100–2000                                        |
| gas–liquid microchannel                    | 3400–18000                                      |

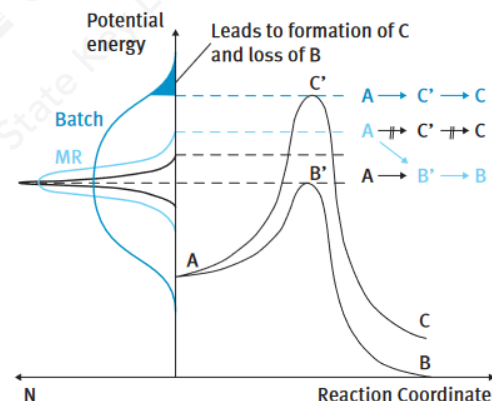
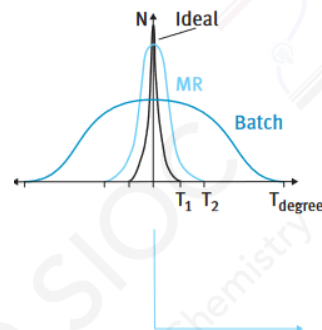
<sup>a</sup>Reproduced from Mallia et al.<sup>59</sup> <sup>b</sup>Calculated for half-filled round-bottom flasks when the liquid is static using,  $\frac{a}{v_{1/2}} = \frac{150}{\sqrt[3]{3v/4\pi}}$ .

### Fast mass transfer: flow (ms to s) vs. batch (min to h)



Heat transfer

$$\frac{Q}{t} = kS \frac{\Delta T}{d}$$



### Fast heat exchange: flow (ms to s) vs. batch (min to h)

## 2.3 流动化学的特征和优势

### Large specific surface area ( $d < 1$ mm, large $S$ )

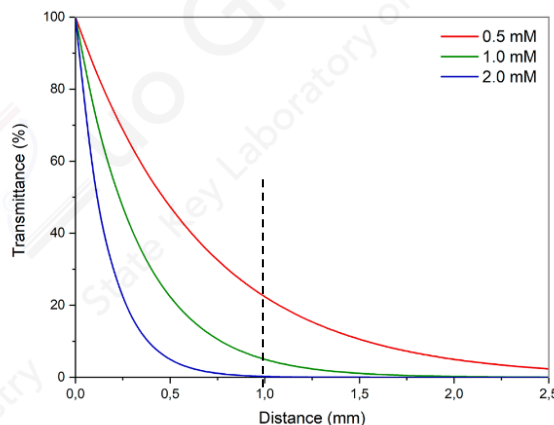


Figure 4. Attenuation of light as a function of distance in a photocatalytic reaction using  $\text{Ru}(\text{bpy})_3\text{Cl}_2$  ( $c = 0.5, 1,$  and  $2$  mM,  $\epsilon = 11000 \text{ cm}^{-1}\cdot\text{M}^{-1}$ ) utilizing the Bouguer–Lambert–Beer correlation

### Sufficient and uniform light

Precise residence time control

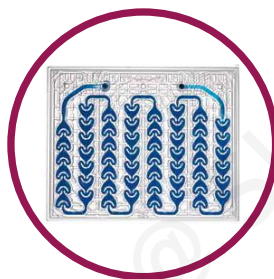
Improved irradiation

Efficient mixing

Improved selectivity and reproducibility

Fast mass and heat exchange

Reliable scale-up



# 提纲

一

可见光催化的重要性和局限性

二

流动化学的基础知识



三

常见的泵和连续流光反应器

四

连续流条件下铈-光催化反应的放大应用研究

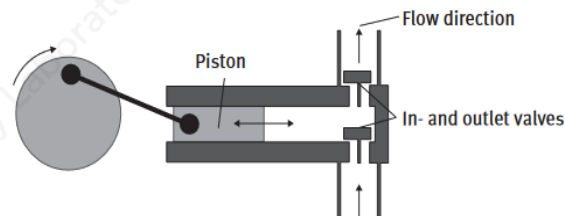
## 3.1 常见的流动化学泵

### ■ syringe pump



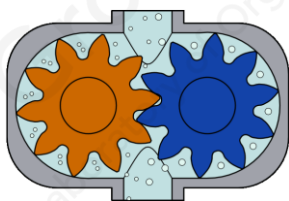
- Precisely deliver constant & pulsation-free fluids, ranging from  $\mu\text{L}/\text{min}$  to  $\text{mL}/\text{min}$
- Constrained capacity and inability to mix

### ■ Piston pump



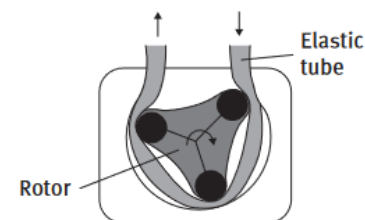
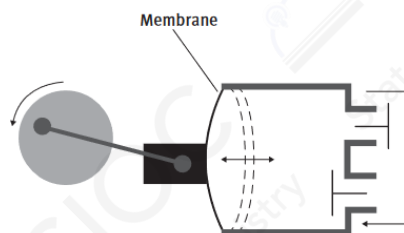
- Most commonly used pumps in flow
- Unsuitable for extremely low flow rates or gas-liquid & liquid-solid mixtures

### ■ Gear pump



- Working under up to 200 bar, suitable for high-viscosity fluids
- Considerable noise and exhibit significant flow pulsation

### ■ diaphragm pump & Peristaltic pump

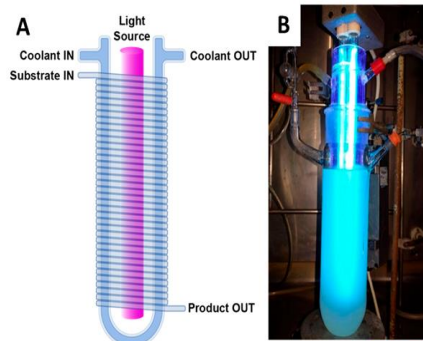


- Suitable for corrosive liquids & liquid-solid mixtures
- Substantial flow pulsation and operate at relatively low pressures (<10 bar)



## 2.2 常见的连续流光反应器

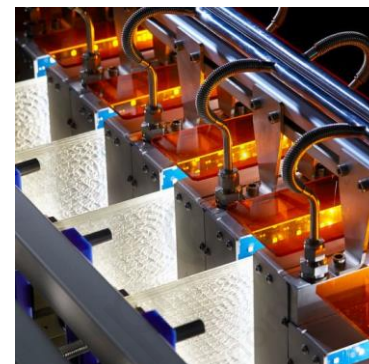
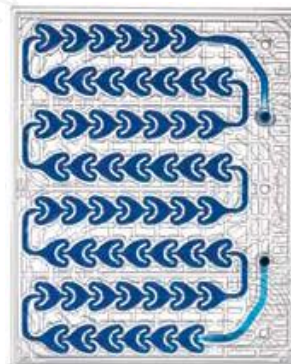
### Capillary microreactors



Flexible & Cheap

Non-durable

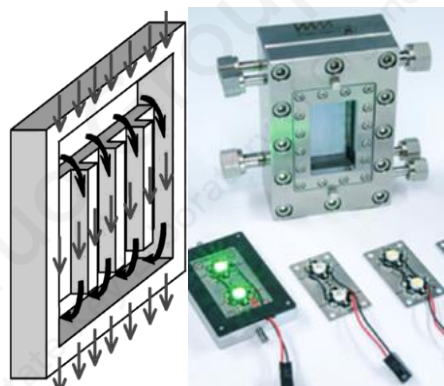
### Glass microreactors



Very large S/V ratio

high cost

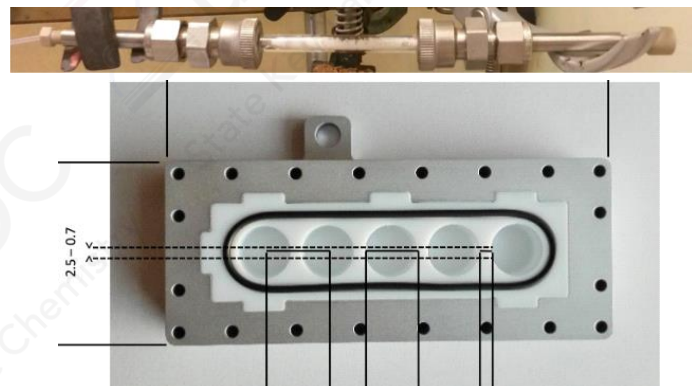
### FFMR and RS-SDR reactor



Very large S/V ratio

Slow flow rate

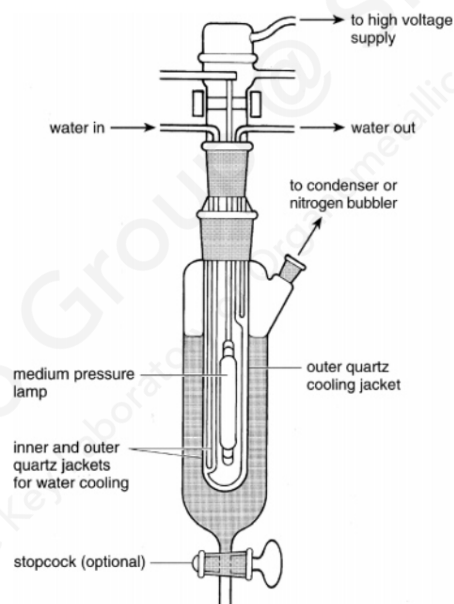
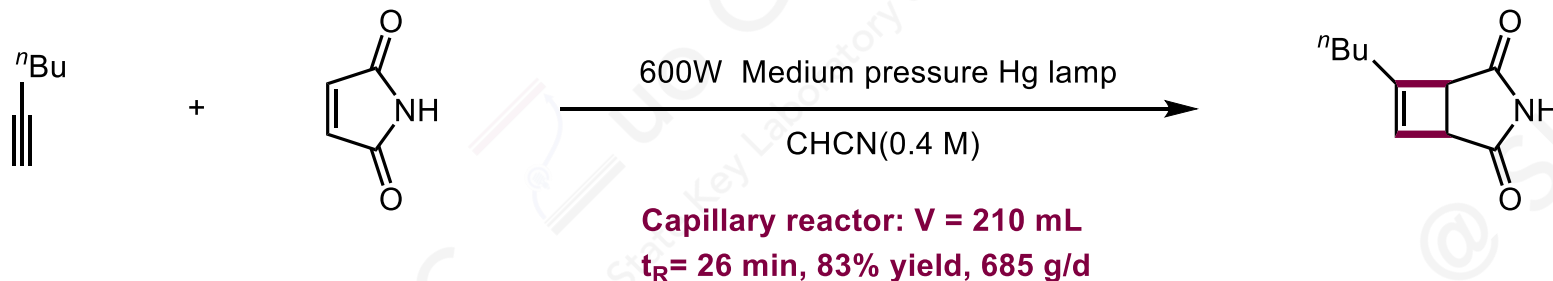
### Packed-bed and CSTR reactor



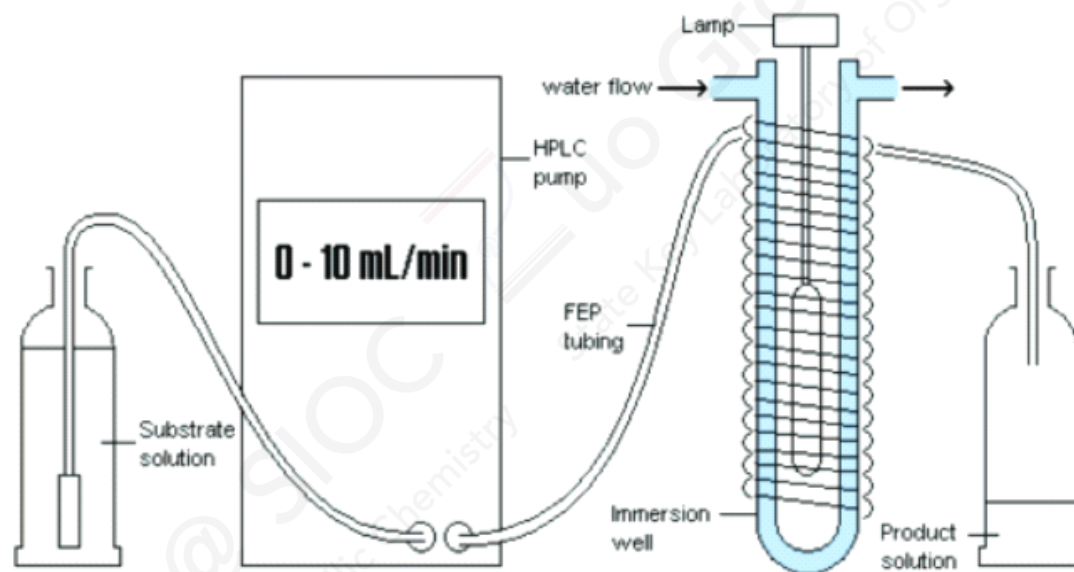
Solid operation

Catalyst leaching

### 3.3 盘管反应器



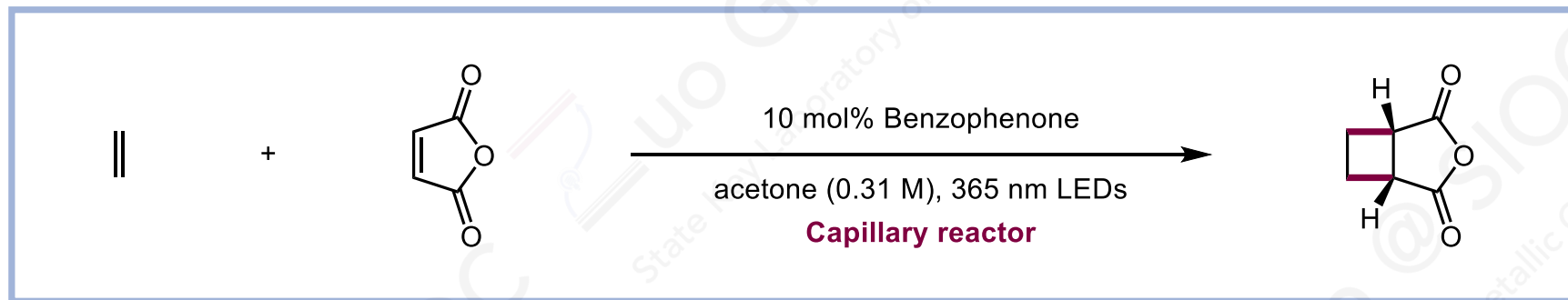
batch photochemical reactor



FEP Capillary reactor



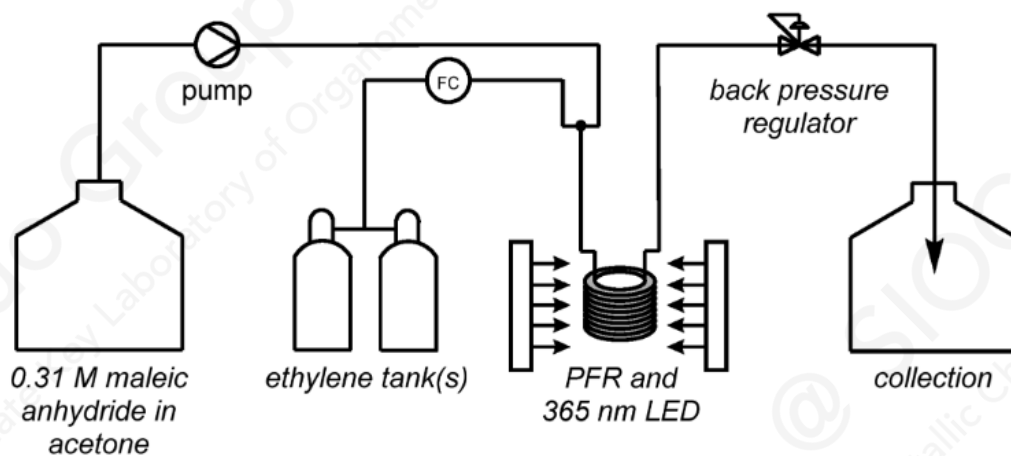
### 3.3 盘管反应器



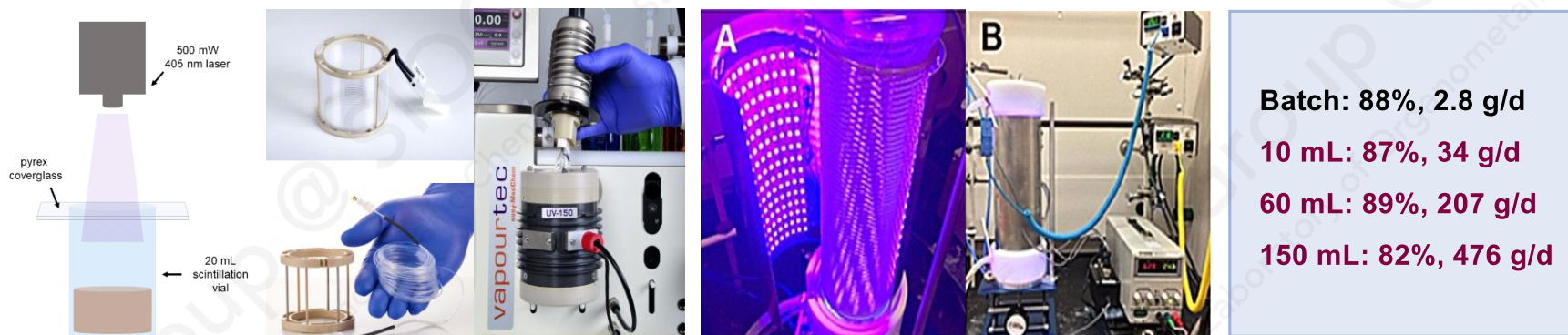
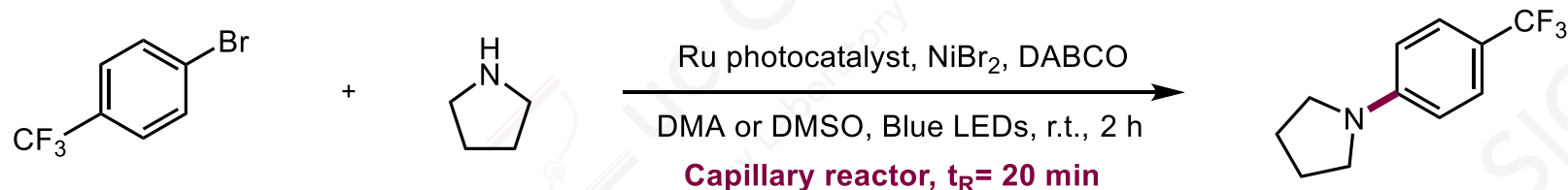
$V = 0.25 \text{ L}$ , 96% yield, 14 g/d

$V = 1.6 \text{ L}$  (32 m),  $t_R = 2 \text{ h}$ ,  
91% yield, 646 g/d

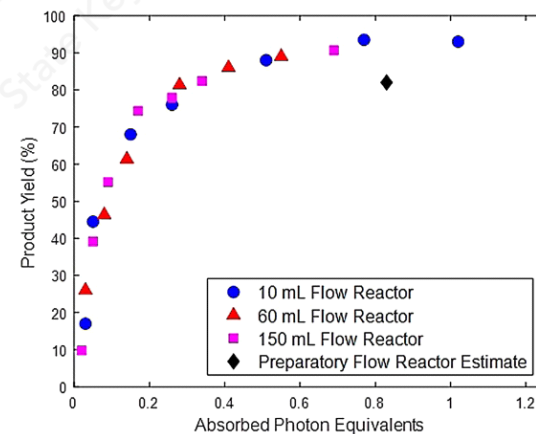
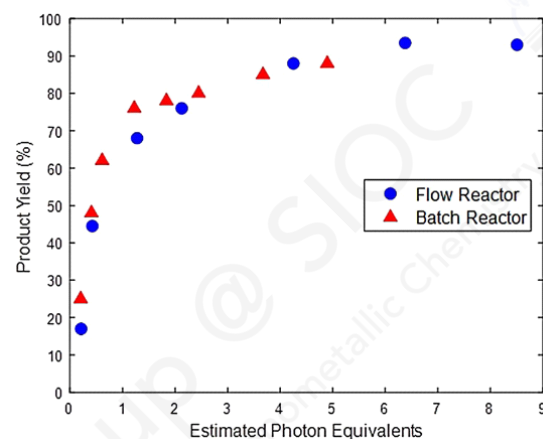
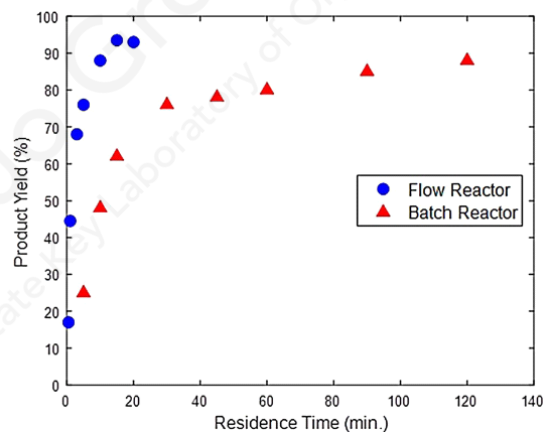
$V = 21 \text{ L}$  (400 m),  $t_R = 2 \text{ h}$ , 61% yield,  
51.8 kg/7d (7.4 kg/d)



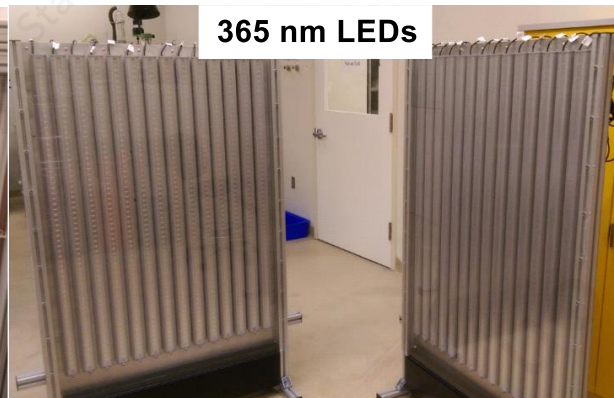
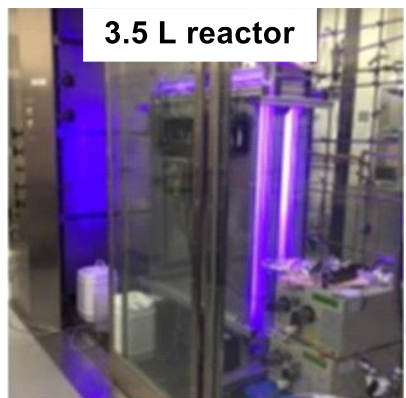
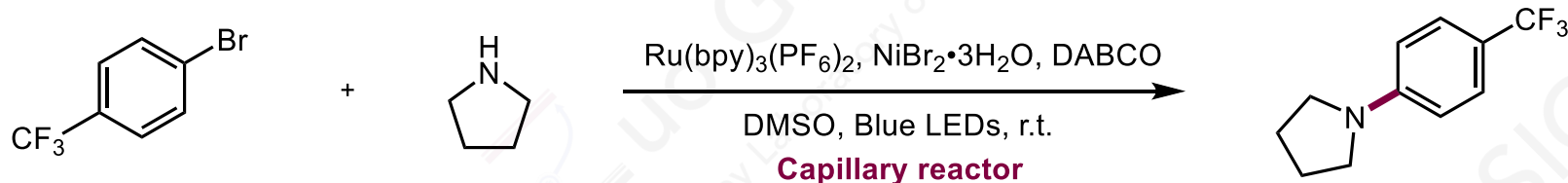
### 3.3 盘管反应器



**Batch: 88%, 2.8 g/d**  
**10 mL: 87%, 34 g/d**  
**60 mL: 89%, 207 g/d**  
**150 mL: 82%, 476 g/d**

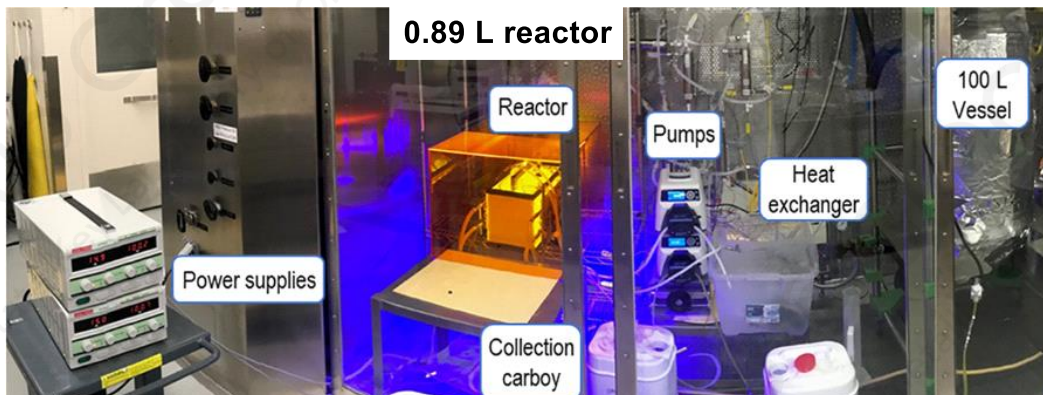


### 3.3 盘管反应器



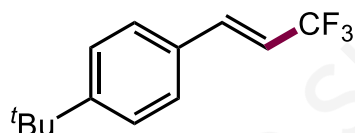
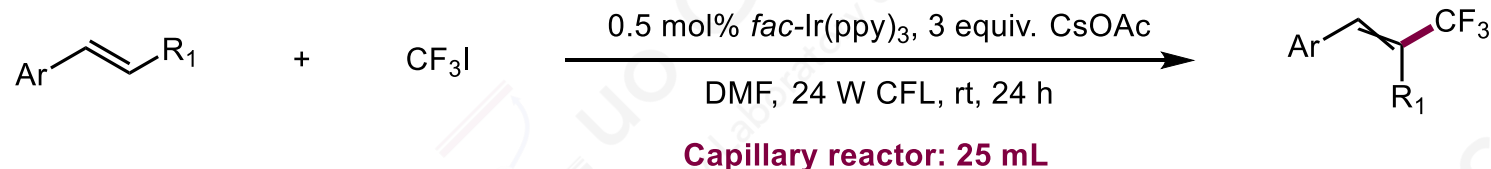
**3.5 L Capillary reactor:**  
 $t_R = 20$  min, 82%, 10 kg/d  
(90% predicted)

**0.89 L Capillary reactor:**  
 $t_R = 8.5$  min, 90%, 12 kg/d



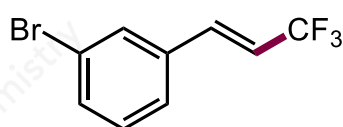
E. B. Corcoran, J. P. McMullen, et al. *Angew. Chem. Int. Ed.* **2020**, 59, 11964–11968.  
F. Lévesque, M. J. Di Maso, et al. *Org. Process Res. Dev.* **2020**, 24, 2935–2940.

### 3.3 盘管反应器



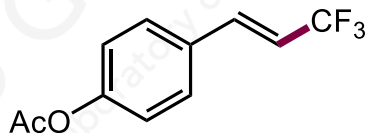
Batch: 92% (45:55 E/Z)

Flow: 97% (94:6 E/Z),  $t_R=30$  min



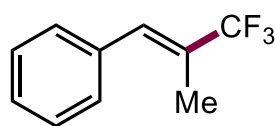
Batch: 81% (46:54 E/Z)

Flow: 80% (97:3 E/Z),  $t_R=90$  min



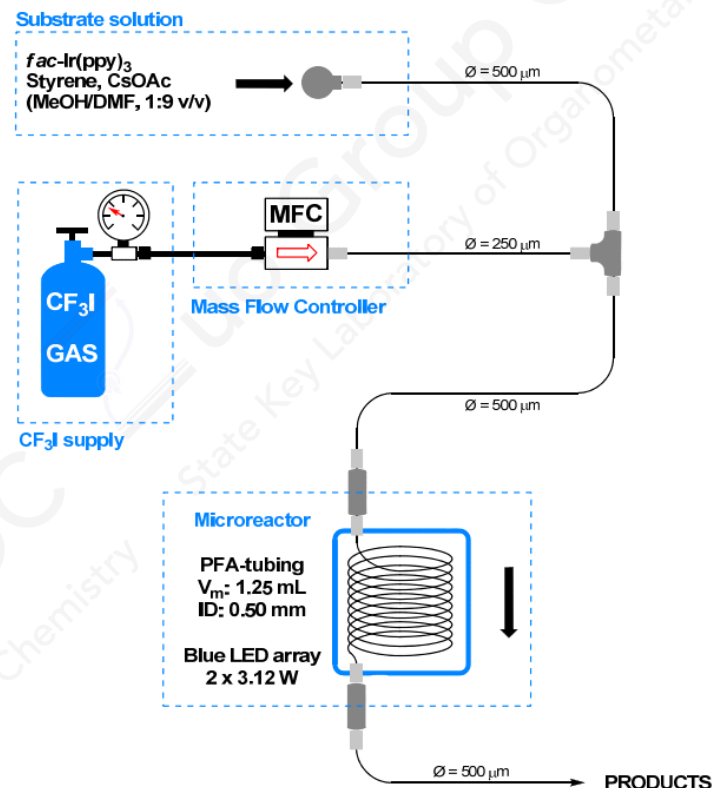
Batch: 91% (19:81 E/Z)

Flow: 98% (98:2 E/Z),  $t_R=90$  min



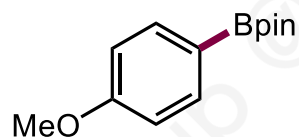
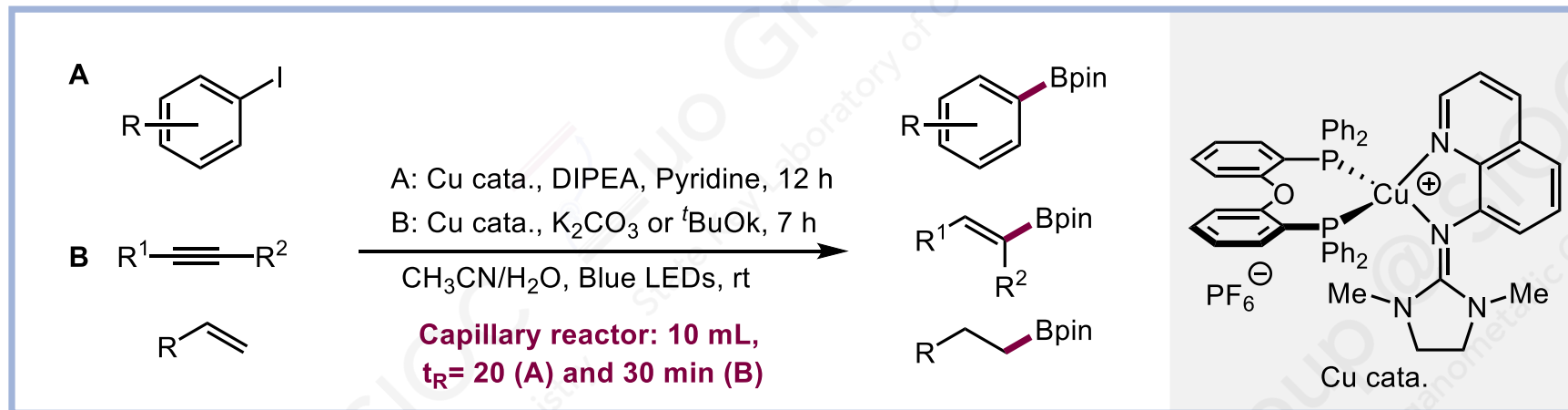
Batch: 75% (69:31 E/Z)

Flow: 76% (E only),  $t_R=80$  min





### 3.3 盘管反应器



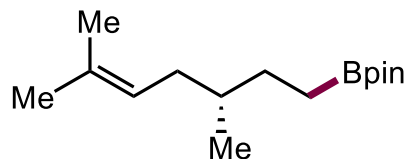
**A: batch: 79%, 0.01 mmol/h  
 flow: 73% yield, 2.2 mmol/h**



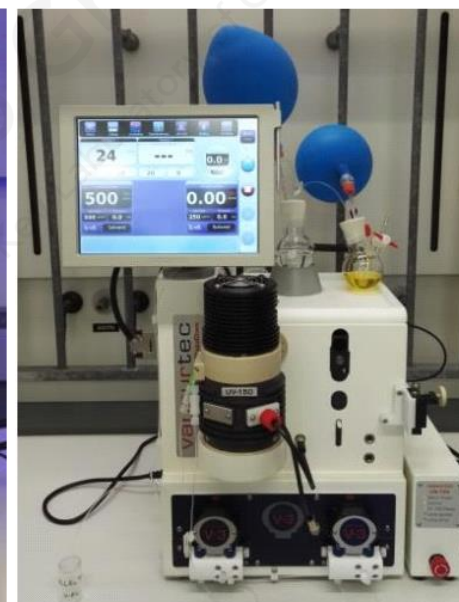
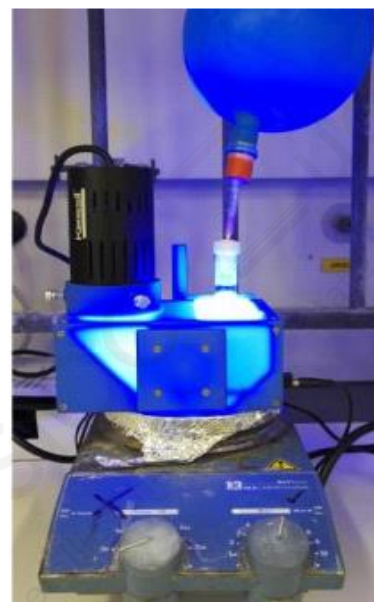
**B: batch: 87%, 0.02 mmol/h  
 flow: 91%, 1.8 mmol/h**



**B: batch: 87%, 0.02 mmol/h  
 flow: 91%, 1.8 mmol/h**

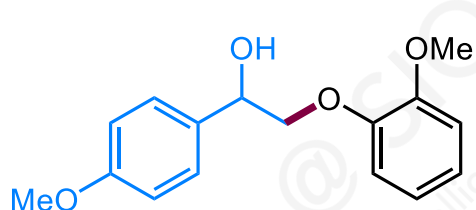
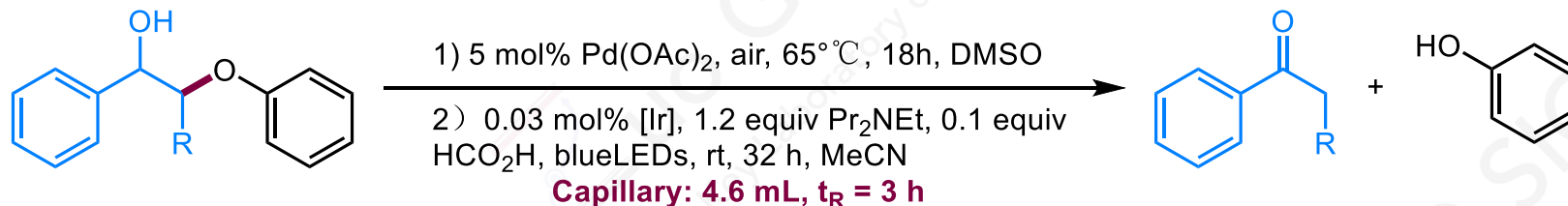


**B: batch: 79%, 0.02 mmol/h  
 flow: 85%, 1.7 mmol/h**

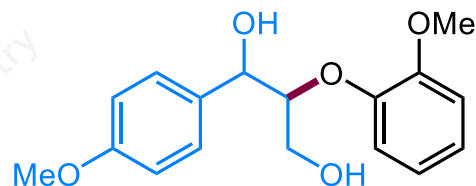


T. Poisson, et al. *Chem. Eur. J.* **2019**, 25, 3262–3266.  
 T. Poisson, et al. *Angew. Chem. Int. Ed.* **2021**, 60, 14498–14503.

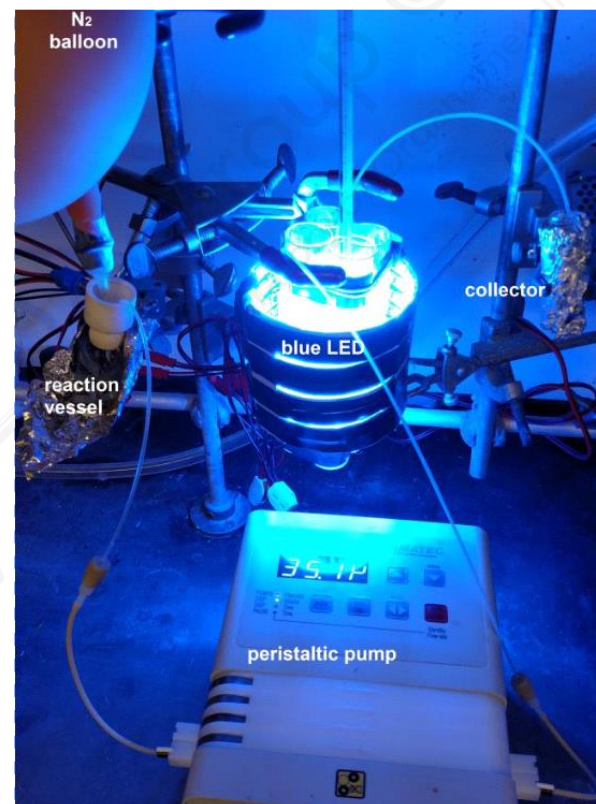
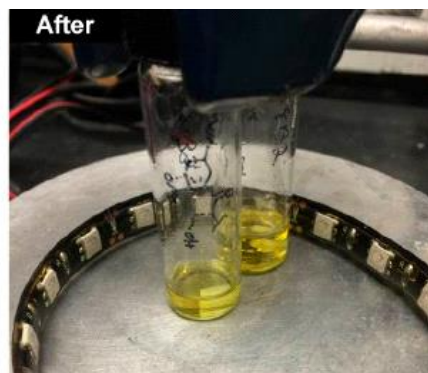
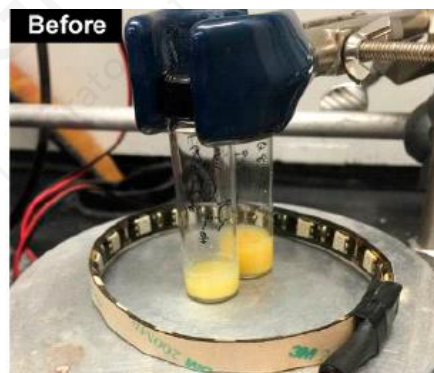
### 3.3 盘管反应器



batch: 94%, 92%, **0.013 mmol/h**  
**flow: 79%, 66%, 0.4 mmol/h**

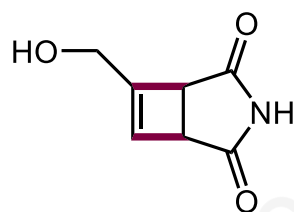
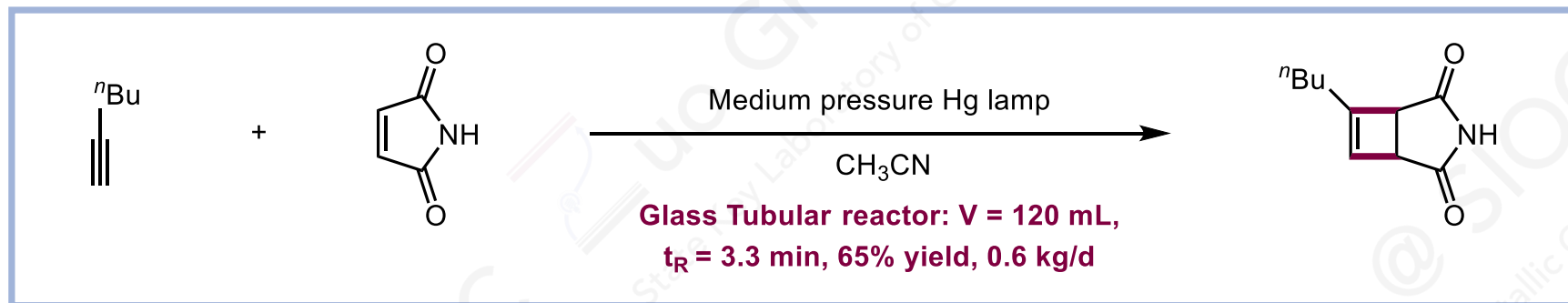


batch: 87%, 83%, **0.013 mmol/h**  
**flow: 77%, 63%, 0.4 mmol/h**

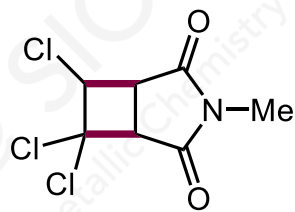


C. R. J. Stephenson, et al. *ACS Catal.* **2019**, 9, 2252–2260.

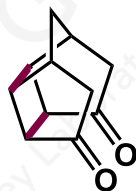
## 3.4 平行管反应器



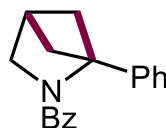
$t_R = 3.3$  min  
 64%, 0.51 kg·d<sup>-1</sup>



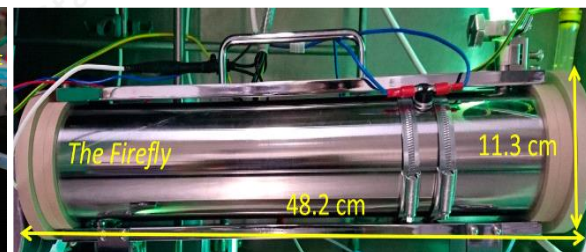
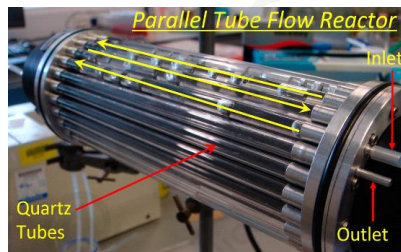
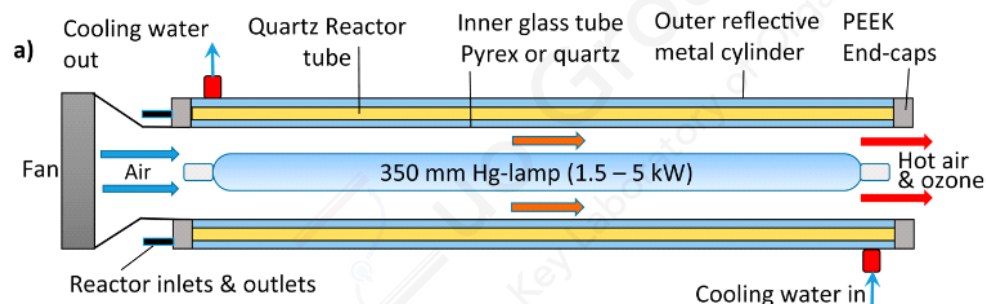
$t_R = 8$  min  
 64%, 0.69 kg·d<sup>-1</sup>



$t_R = 3.3$  min  
 89%, 8.1 kg·d<sup>-1</sup>

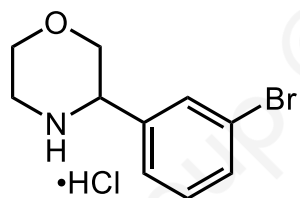
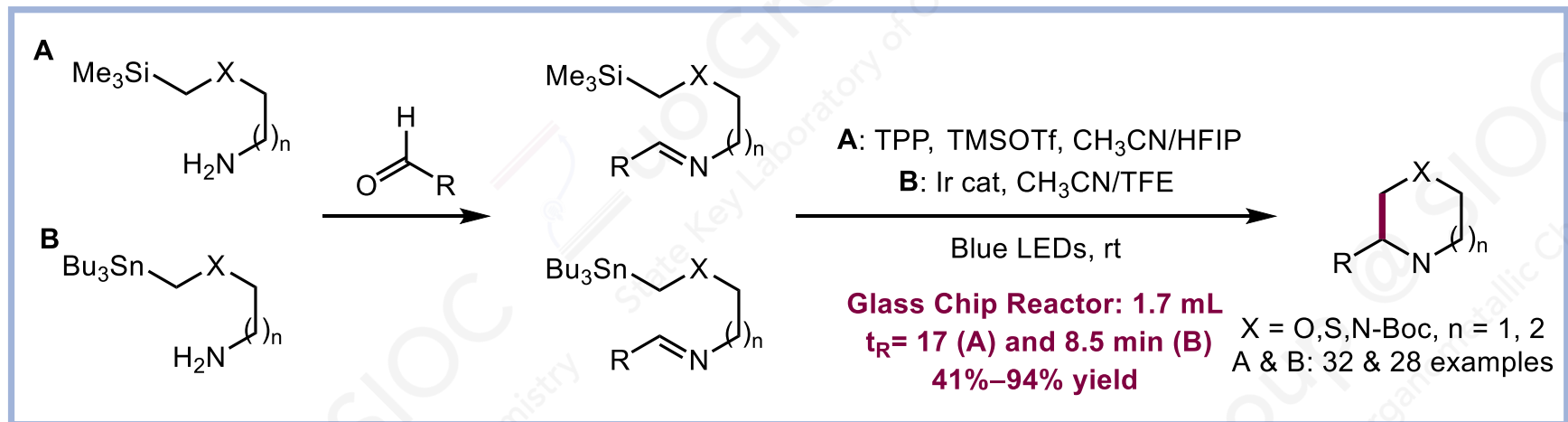


$t_R = 13$  min  
 86%, 1.2 kg·d<sup>-1</sup>

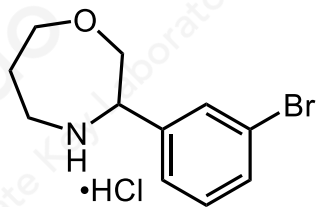




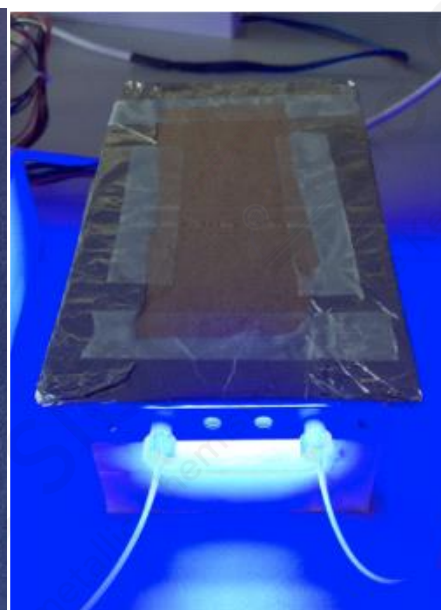
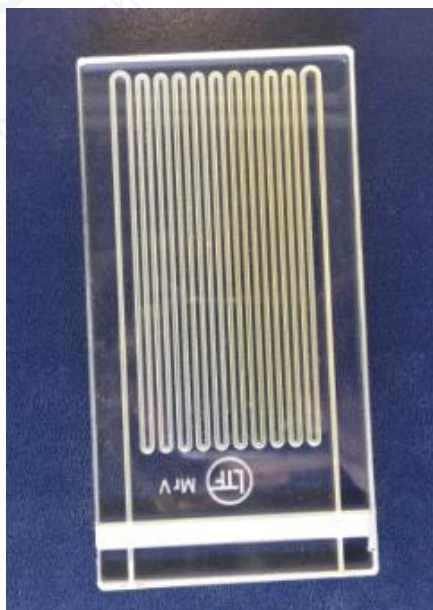
## 3.5 芯片反应器



**A: 67% yield, 2.8 g/d**

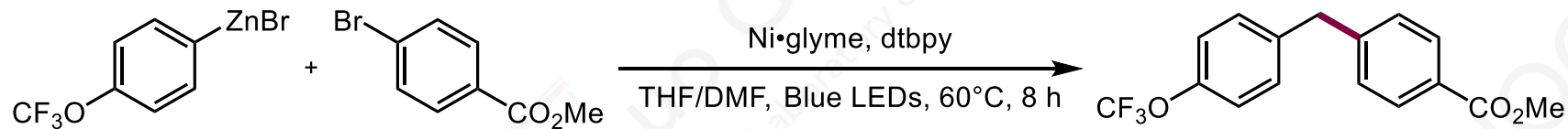


**B: 41% yield, 1.7 g/d**



J. W. Bode, et al. *Org. Lett.* **2017**, 19, 4696–4699.  
 J. W. Bode, et al. *Org. Lett.* **2018**, 20, 2071–2075.

### 3.5 芯片反应器



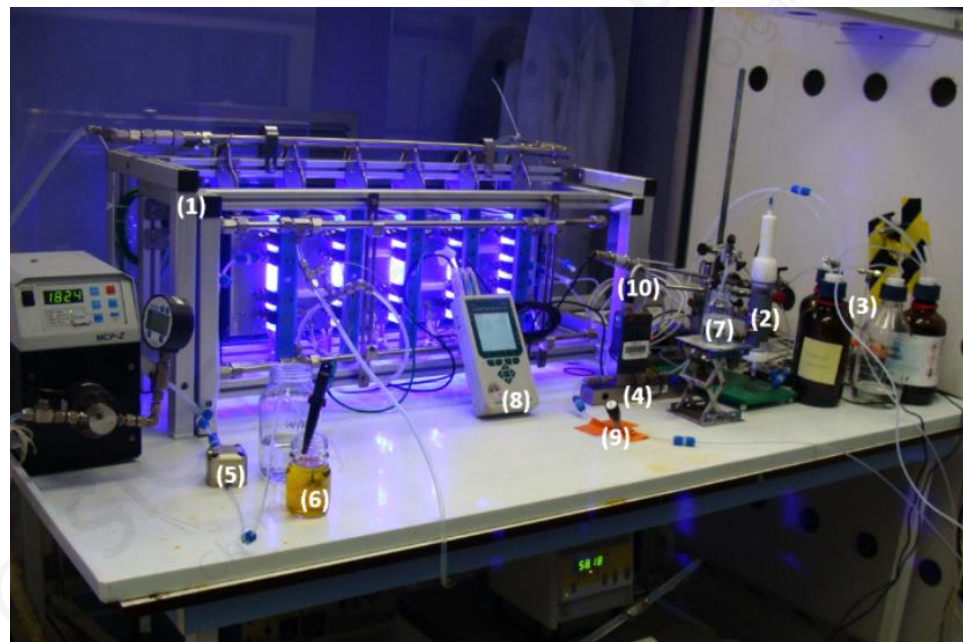
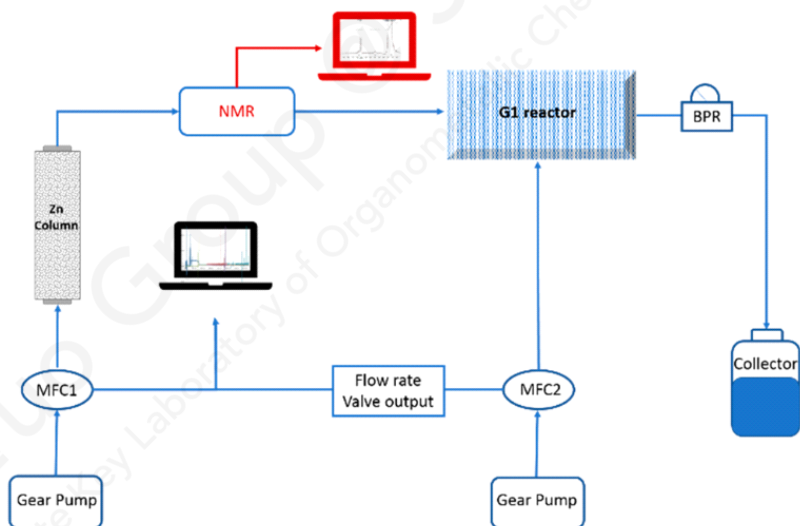
**Batch: 39%**

**1. Capillary reactor: 10 mL,  $t_R$  = 20 min**

**Flow 1: 93%, 0.8 g/h**

**2. Glass reactor: 40 mL,  $t_R$  = 15 min**

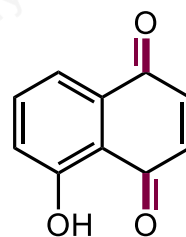
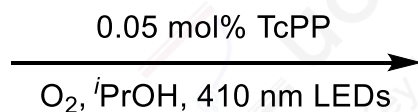
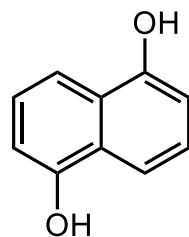
**Flow 2: 93%, 5.6 g/h**



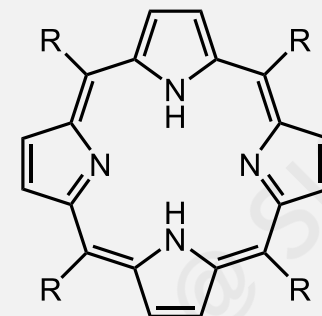
J. Alcázar, et al. *Angew. Chem. Int. Ed.* **2018**, 57, 8473–8477.

J. Alcazar, et al. *J. Org. Chem.* **2019**, 84, 4748–4753.

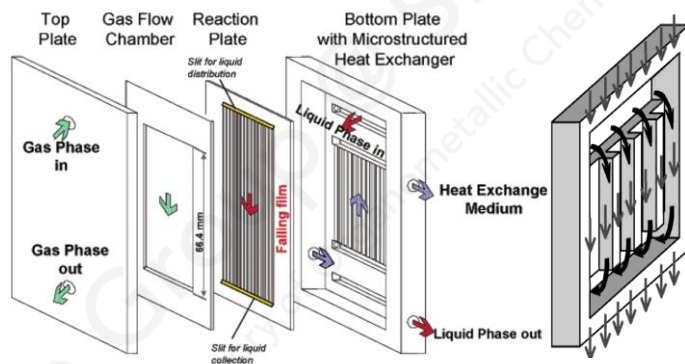
## 3.6 降膜反应器



**FFMR reactor: 54x0.6x0.2 mm, 52  $\mu\text{L}$**   
 **$t_R = 20 \text{ s}$ , 96% yield, 0.36 g/d**



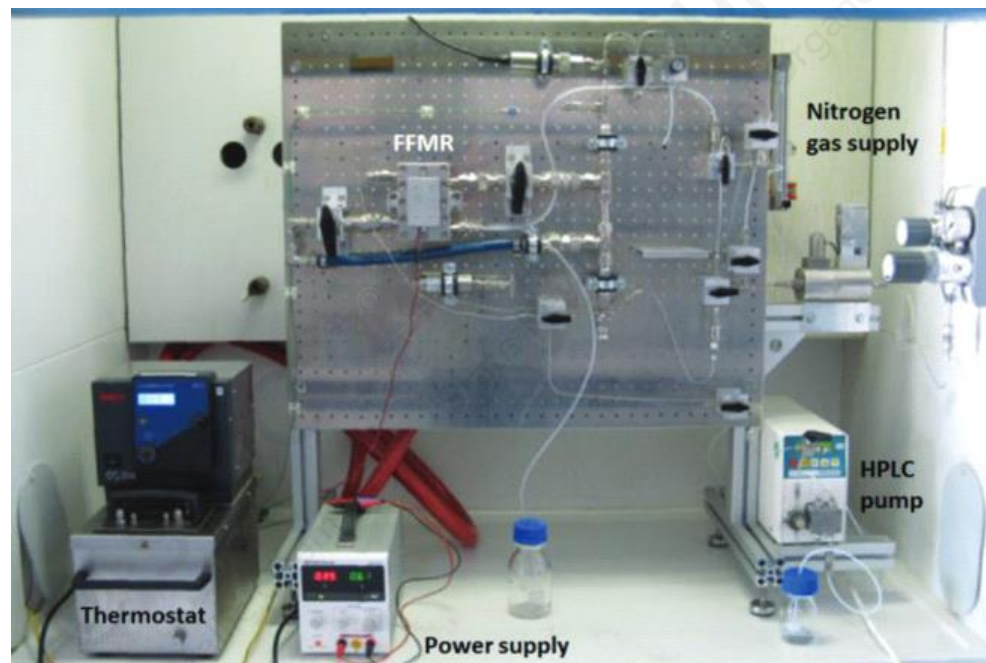
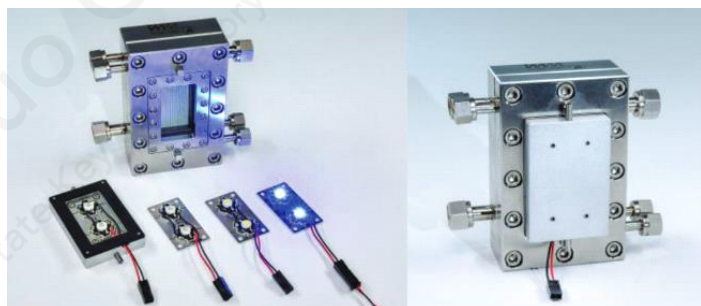
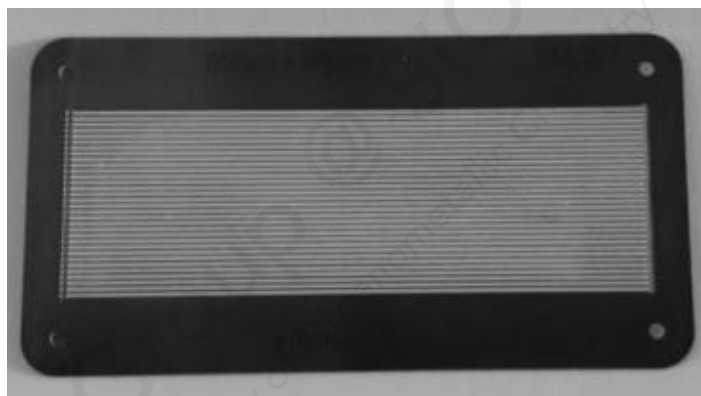
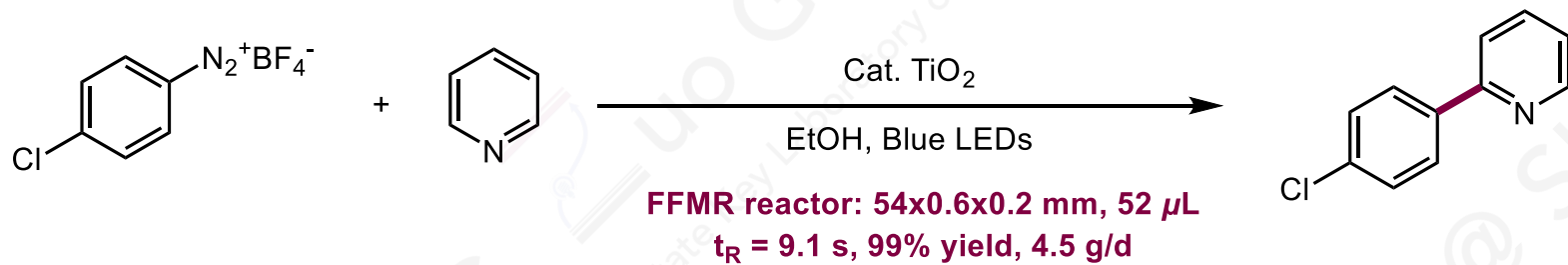
TcPP:  $R = (p\text{-CO}_2\text{H})\text{C}_6\text{H}_4$



T. H. Rehm, et al. *React. Chem. Eng.* **2016**, 1, 636–648.

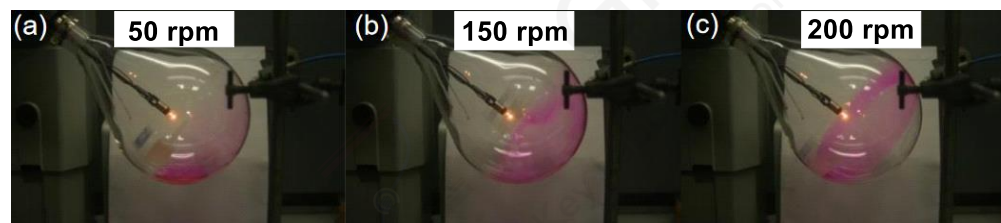
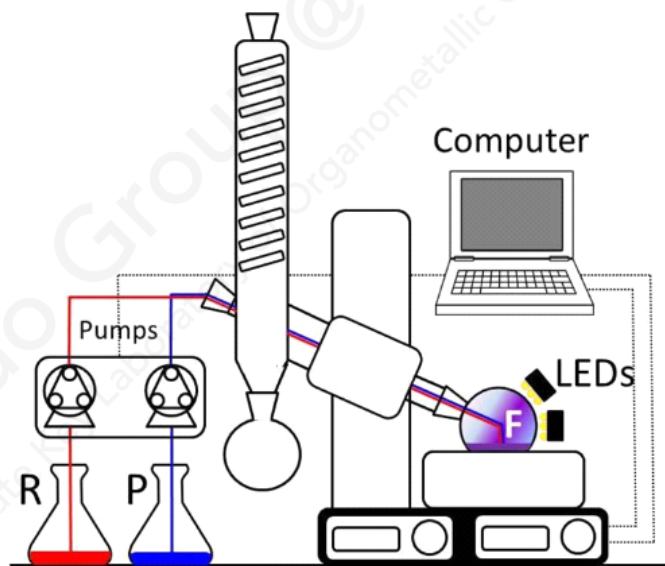
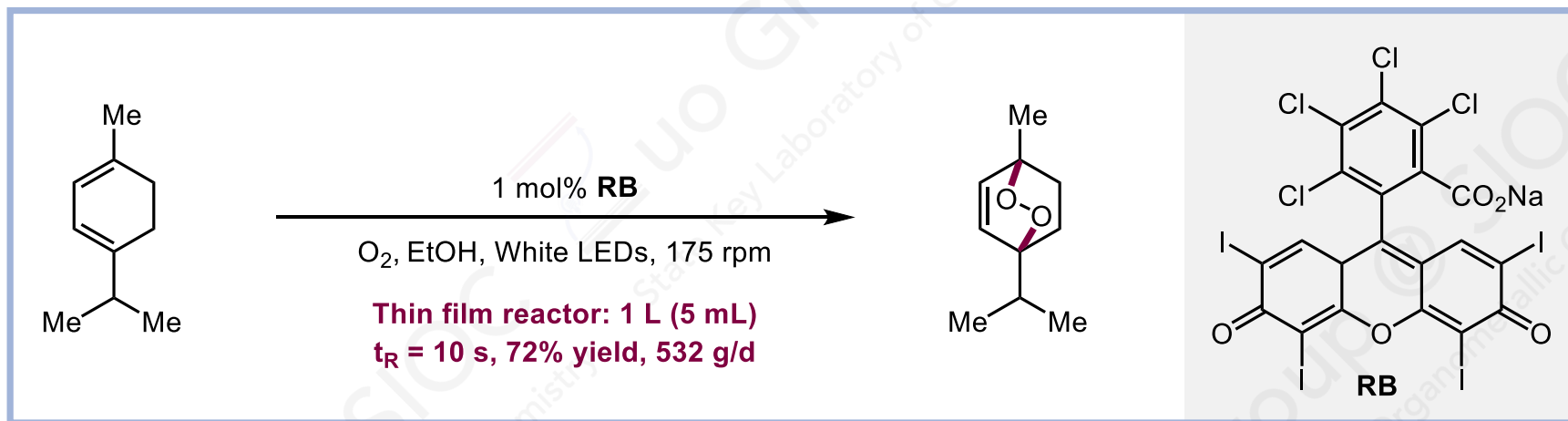


### 3.6 连续流光反应器：降膜反应器

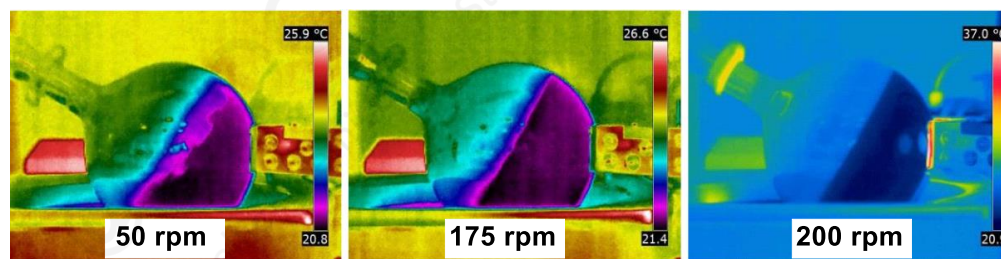


M. Rueping, T. H. Rehm, et al. *Green Chem.* **2017**, 19, 1911–1918.

## 3.7 基于旋蒸的膜反应器

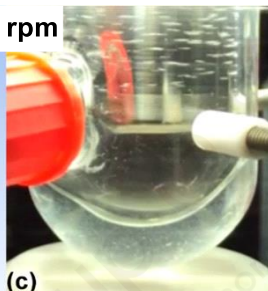
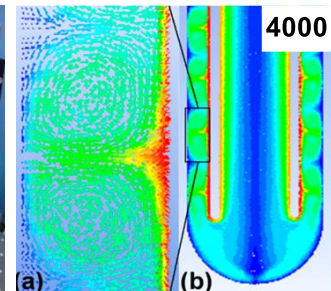
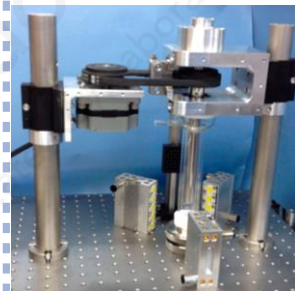
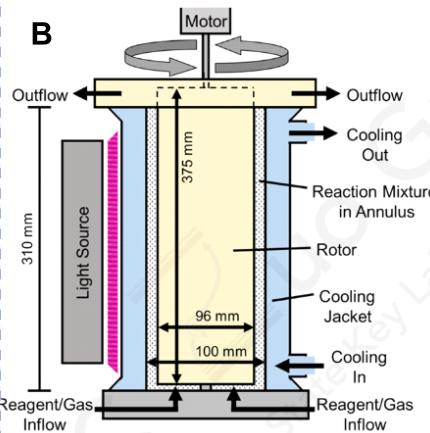
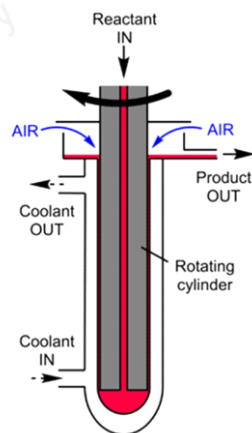
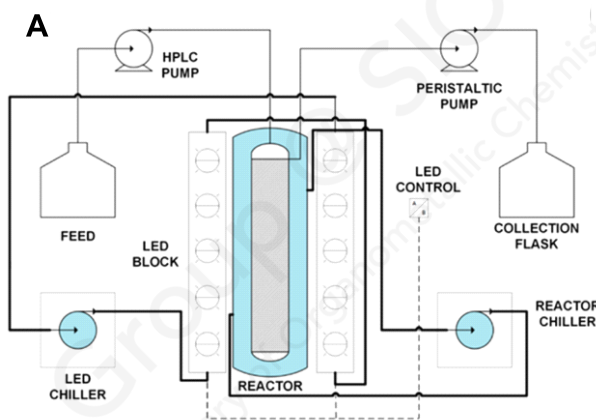
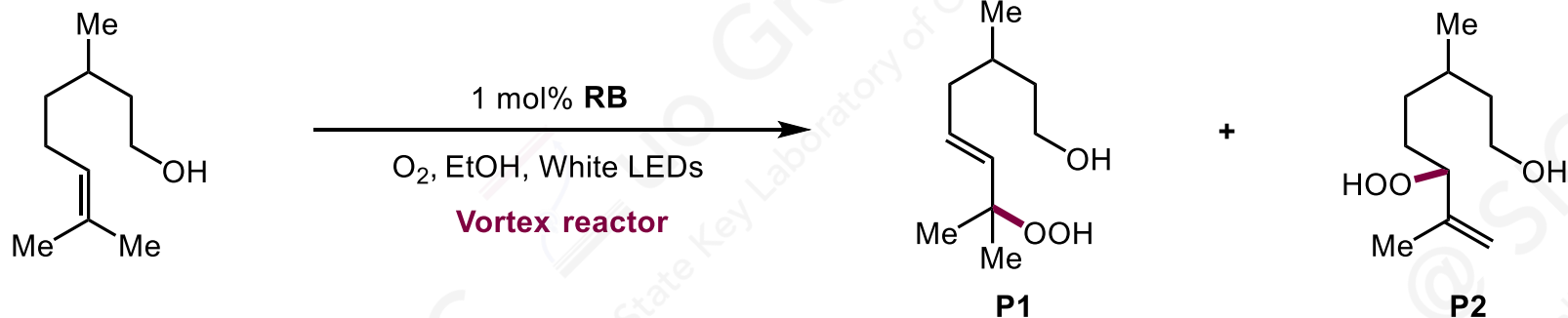


film thickness measurements showing the three types of fluid behaviour



Thermal imaging photographs of a 1 L RB flask in a water bath at r.t..

# 3.8 涡流反应器



4000 rpm

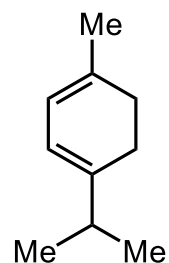


**Vortex reactor A:** ID = 1 mm, V = 8 mL, rpm = 4000,  $t_R$  = 32 min, 95% yield, 6.4 g/d.

**Vortex reactor B:** ID = 2 mm, V = 280 mL, rpm = 660,  $t_R$  = 7 min, 92% yield, 2.0 g/d.



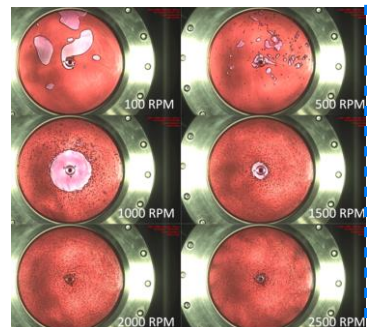
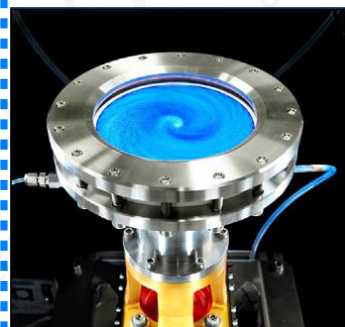
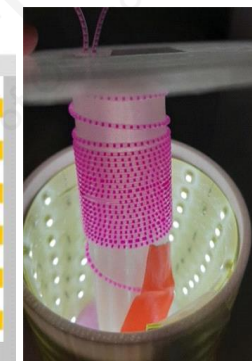
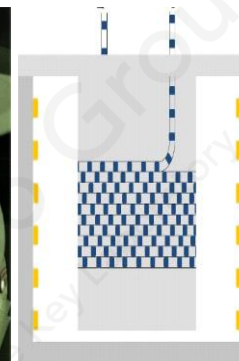
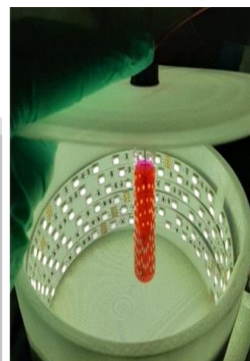
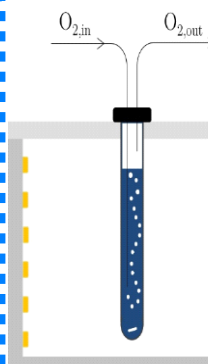
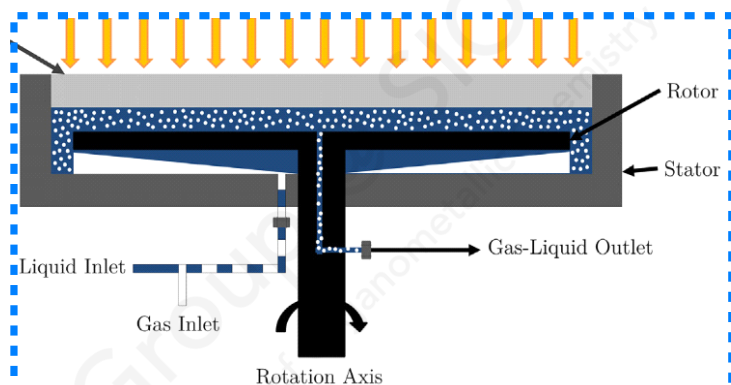
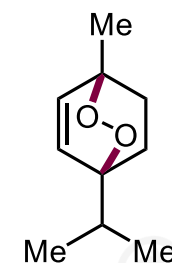
### 3.9 转子-定子圆盘反应器



1 mol% **RB**

O<sub>2</sub>, EtOH, White LEDs

**RotorStator Spinning Disk Reactor (RS-SDR)**

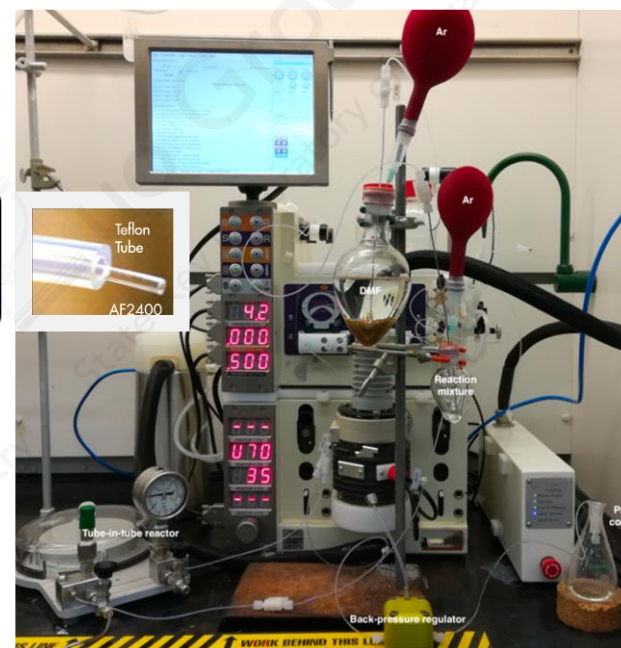
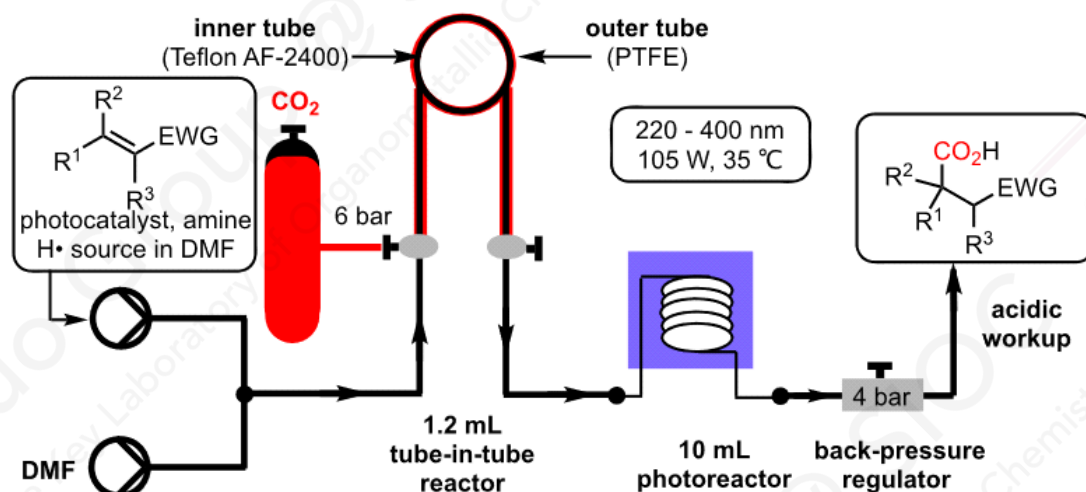
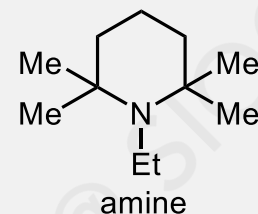
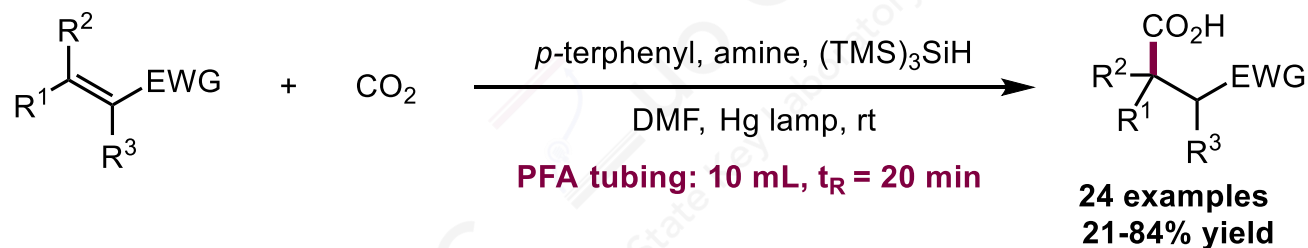


**RS-SDR reactor:  $V = 64$  (27) mL, rpm = 2000,  $t_R = 27$  s, 67% yield, 1.1 kg/d, 2.8 M/s Productivity /Volume**

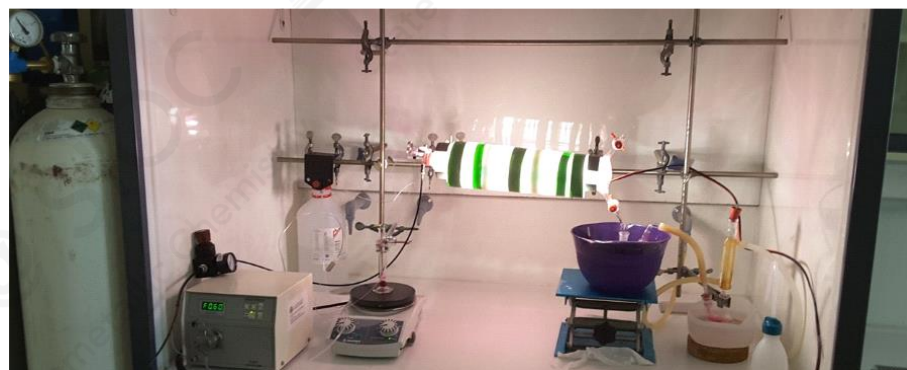
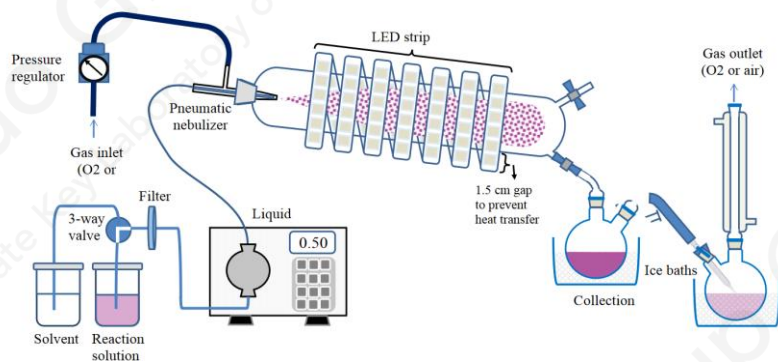
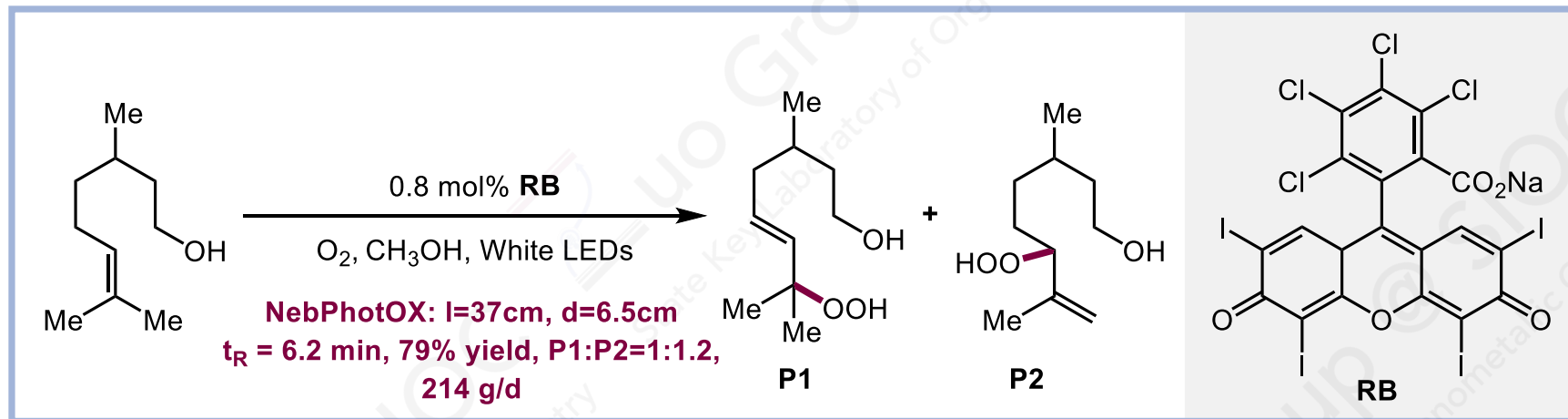
**Batch reactor: 0.021 M/s Productivity /Volume**

**Capillary reactor: 1.2 M/s Productivity /Volume**

## 3.10 管中管反应器

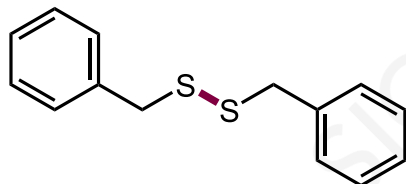
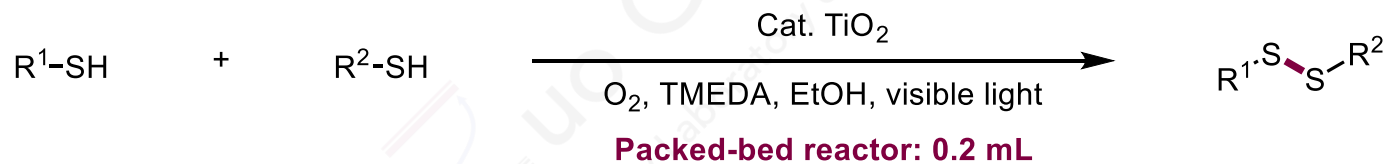


## 3.11 气动雾化反应器

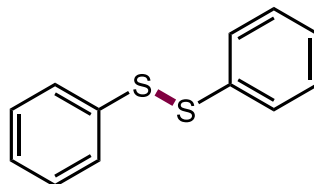


G. Vassilikogiannakis, et al. *ChemPhotoChem* **2017**, 1, 173–177.

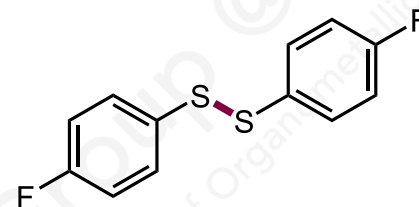
## 3.12 填充床反应器



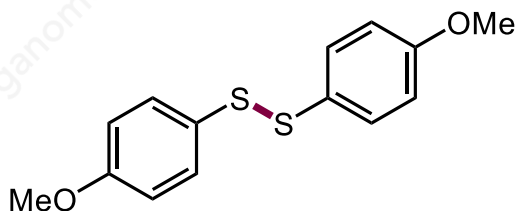
**batch: 8 h, 75%**  
**flow: 5 min, 60%**



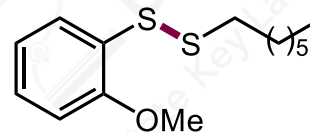
**batch: 2 h, 95%**  
**flow: 3 min, 99%**



**batch: 3 h, 96%**  
**flow: 3 min, 93%**



**batch: 5 h, 91%**  
**flow: 3 min, 78%**



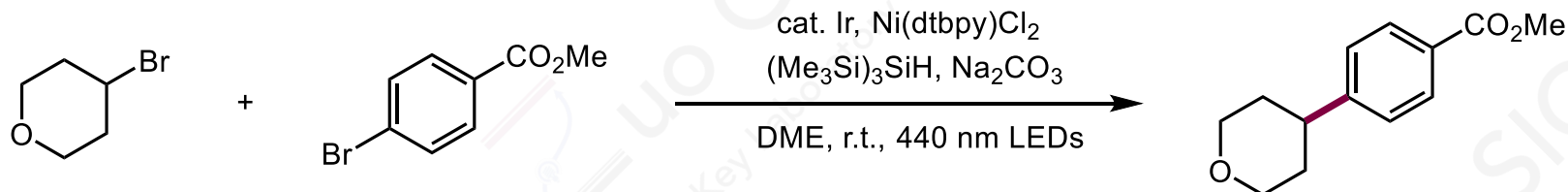
**flow: 3 min, 86%**



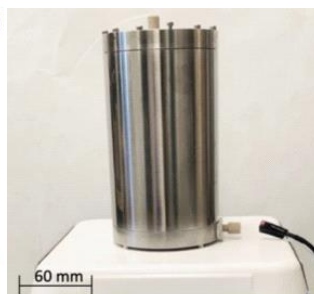
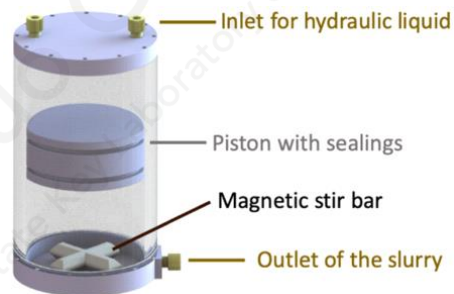
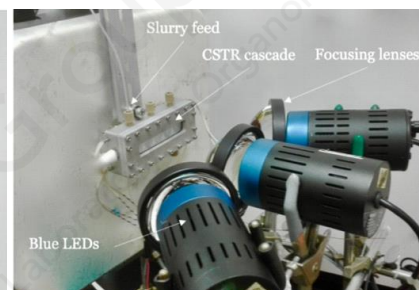
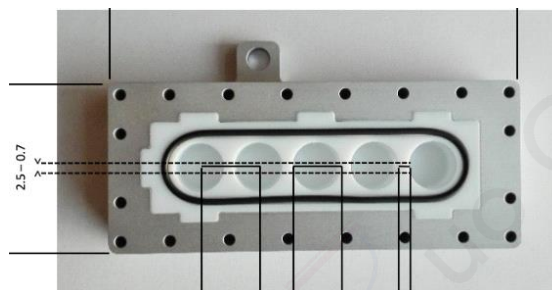
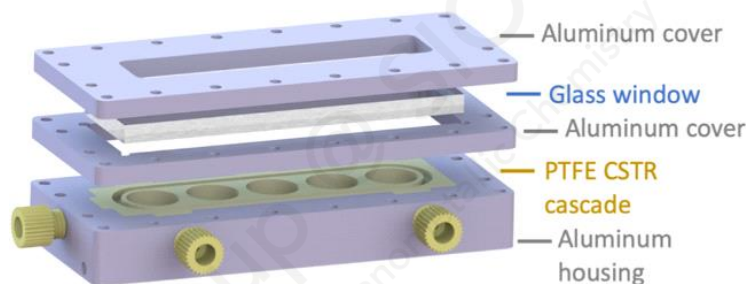
T. Noël, et al. *ChemSusChem* **2016**, 9, 1781–1785.



### 3.13 连续搅拌罐级联反应器



**Continuous Stirred-Tank Reactor (CSTR) Cascade:**  
 5.3 mL,  $t_R$  = 30 min, 77% yield (batch: 80%), 77 mg/h

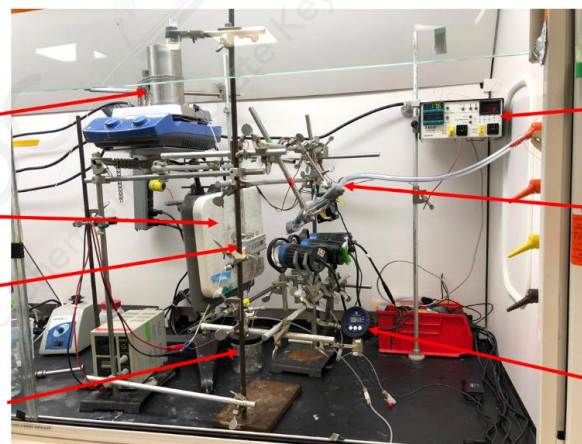


Slurry pump on a stirplate

Magnetic stirrer which controls the stir bars in the CSTR cascade

Photo CSTR

Pressurized fraction collector



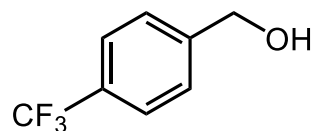
Temperature control

Air cooling

Pressure control

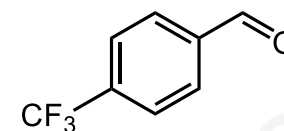


### 3.14 超声辅助盘管反应器

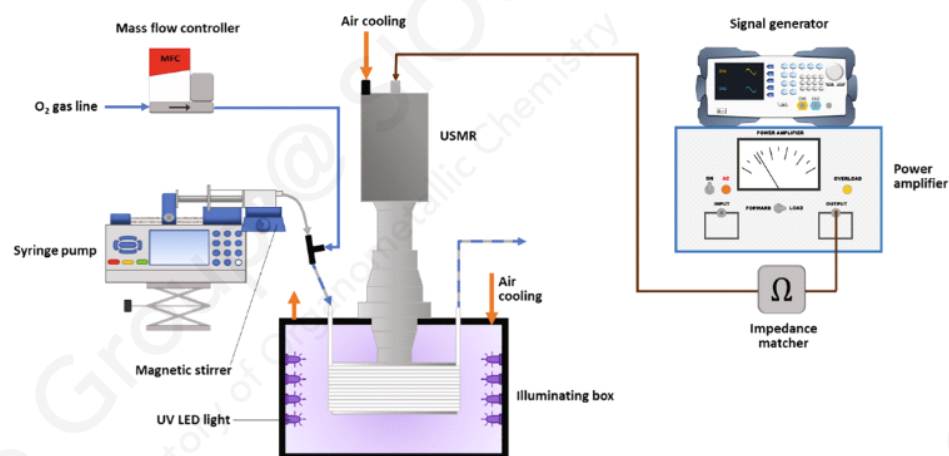


Cat.  $\text{TiO}_2$

$\text{O}_2$ , MeCN, 365 nm LEDs

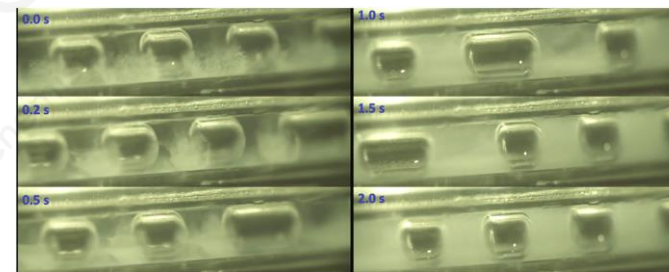
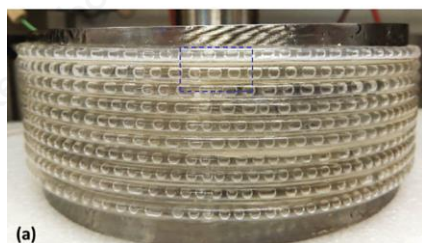


**Ultrasonic Glass Capillary Reactor:**  
 $V = 12.9 \text{ mL}$ ,  $t_R = 28 \text{ min}$ , 86% yield, 3.6 time higher



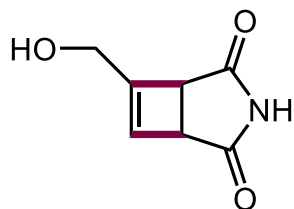
No US

US=60W



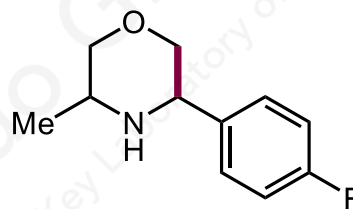
# 3.15 连续流反应器在光催化领域的应用概览

## Photocycloadditions



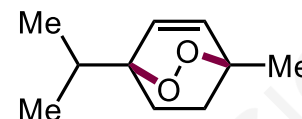
flow: **64%, 3.3 min, 0.5 kg/d**  
batch: 79%, 1 h

## Photocyclization



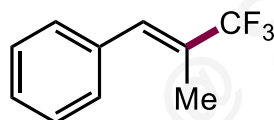
flow: **57%, 8.5 min**  
batch: 59%, 16 h

## Photocatalytic Oxidations

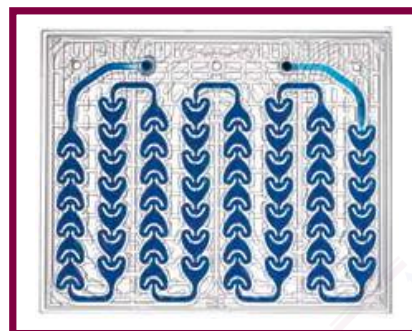


flow: **270 mmol/h**  
batch: 0.4 mmol/h

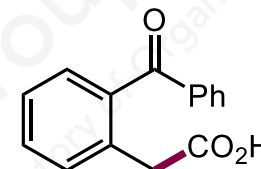
## Photocatalytic Halogenation



flow: **76%, 1.3 h, E only**  
batch: 73%, 24 h, 69:31 E/Z



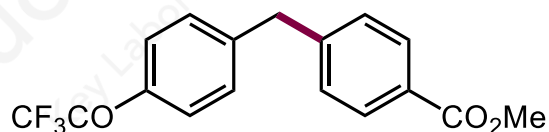
## Photocatalytic Carboxylation



flow: **98%, 1.3 h**  
batch: 75%, 24 h

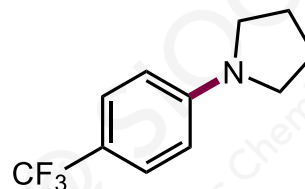
## Photocatalytic Cross-Coupling Reactions

### C—C Formation



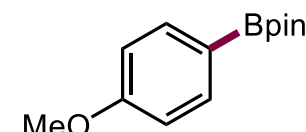
flow: **93%, 15 min, 5.6 g/h**  
batch: 39%, 8 h, 7.7 mg/h

### C—N Formation



flow: **90%, 8.5 min, 12kg/d**  
batch: 96%, 19 h

### C—B Formation



flow: **81%, 20 min**  
batch: 79%, 12 h

# 提纲

一

可见光催化的重要性和局限性

二

流动化学的基础知识

三

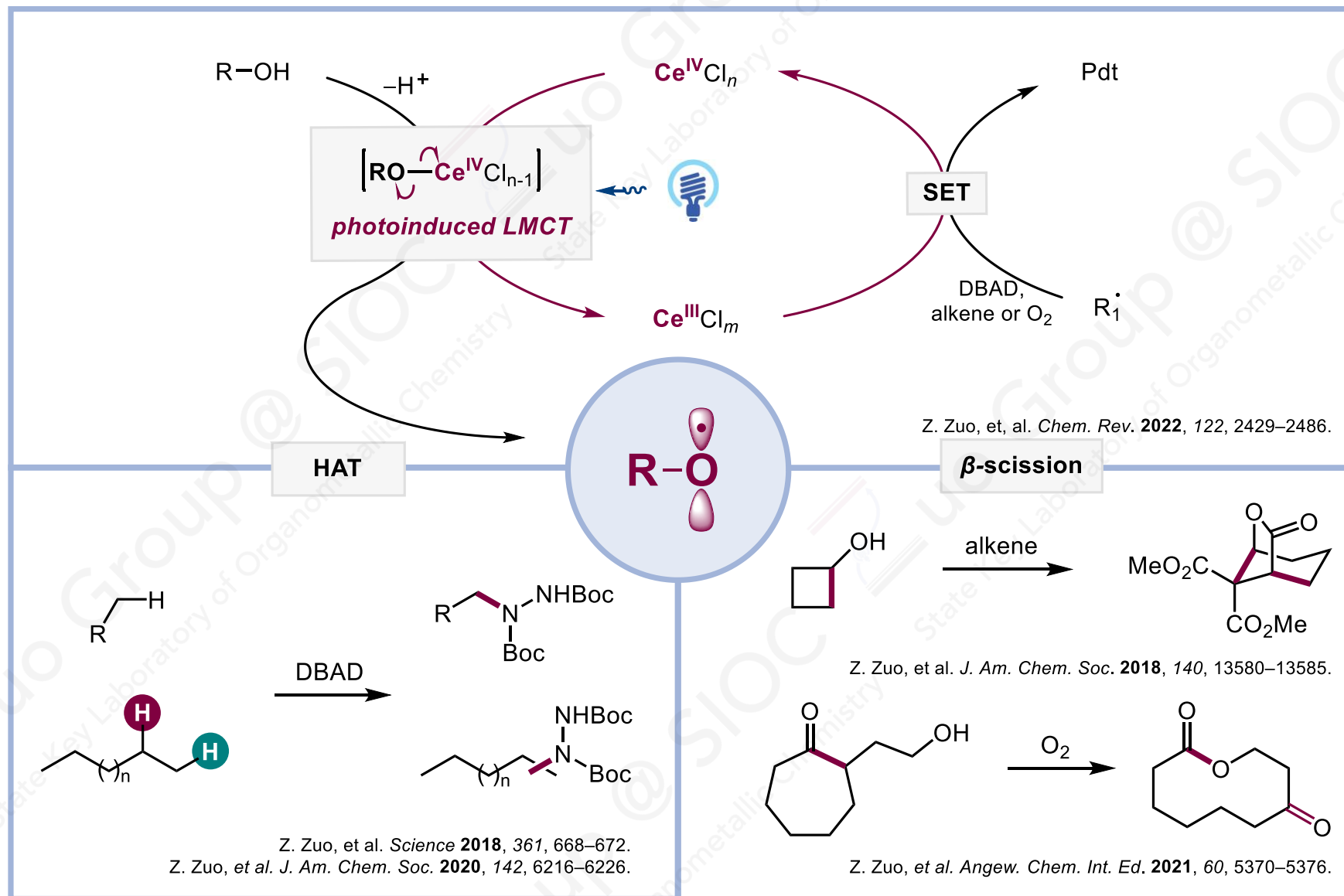
常见的泵和连续流光反应器

四

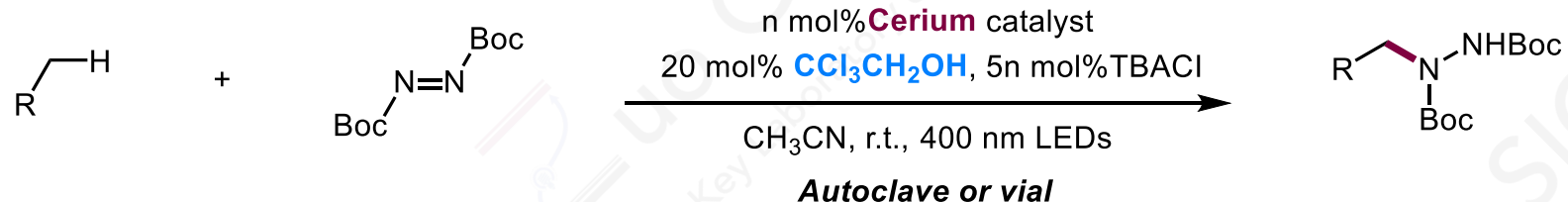
连续流条件下铈-光催化反应的放大应用研究



# 3.1 氧自由基介导的转换



## 3.2 铈-光催化的轻质烷烃胺化反应



| Entry          | Alkane                               | Cerium catalyst                                      | Loading (mol%) | time (h) | Yield (%) | TON  |
|----------------|--------------------------------------|------------------------------------------------------|----------------|----------|-----------|------|
| 1 <sup>a</sup> | Methane (50 bar)                     | (n-Bu <sub>4</sub> N) <sub>2</sub> CeCl <sub>6</sub> | 0.5            | 2        | 41        | 82   |
| 2 <sup>b</sup> | Ethane (10 bar)                      | CeCl <sub>3</sub>                                    | 0.01           | 4        | 97        | 2900 |
| 3 <sup>c</sup> | Propane (1bar)                       | CeCl <sub>3</sub>                                    | 0.5            | 9        | 70        | 140  |
| 4 <sup>d</sup> | Butane (1 bar)                       | CeCl <sub>3</sub>                                    | 0.5            | 6        | 76        | 152  |
| 5              | R = c-C <sub>6</sub> H <sub>11</sub> | CeCl <sub>3</sub>                                    | 0.5            | 16       | 81        | 162  |

<sup>a</sup> no TBACl, with 40 mol% TFA. <sup>b</sup>DIAD instead of DBAD. <sup>c</sup>2°:1°=1:1. <sup>d</sup>2°:1°=1.7:1.

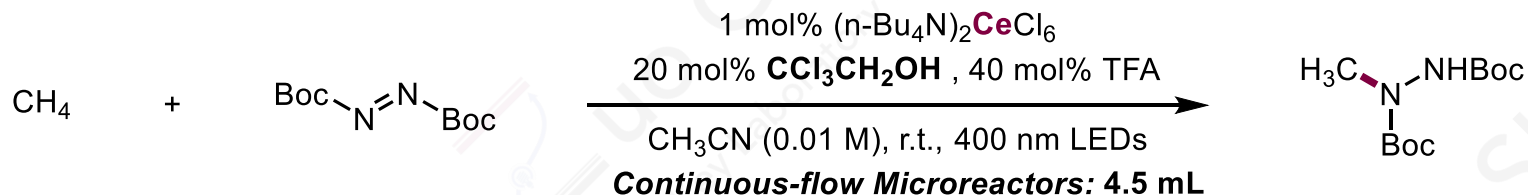
Datas from A. Hu, and J.-J. Guo

Z. Zuo, et al. *Science* **2018**, 361, 668–672.

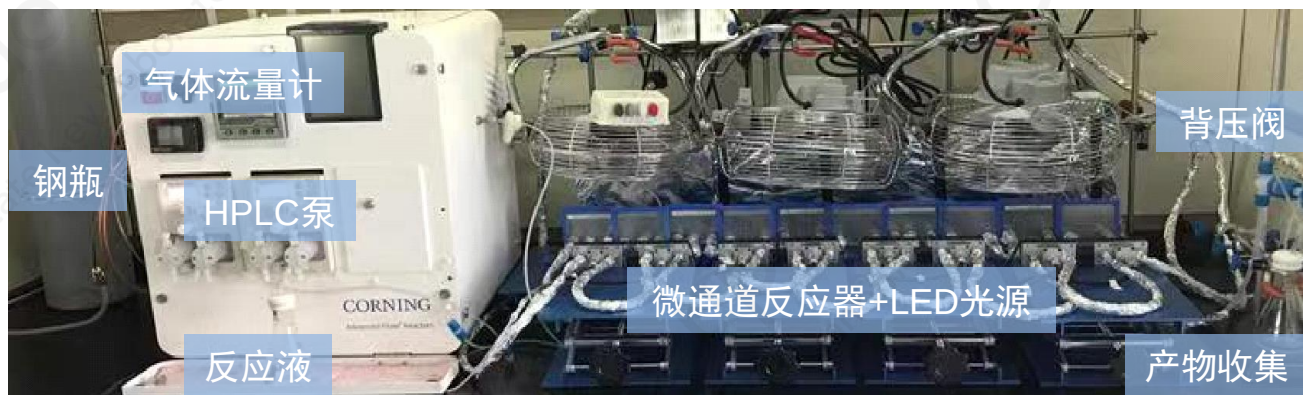




## 3.2 铈-光催化的轻质烷烃胺化反应

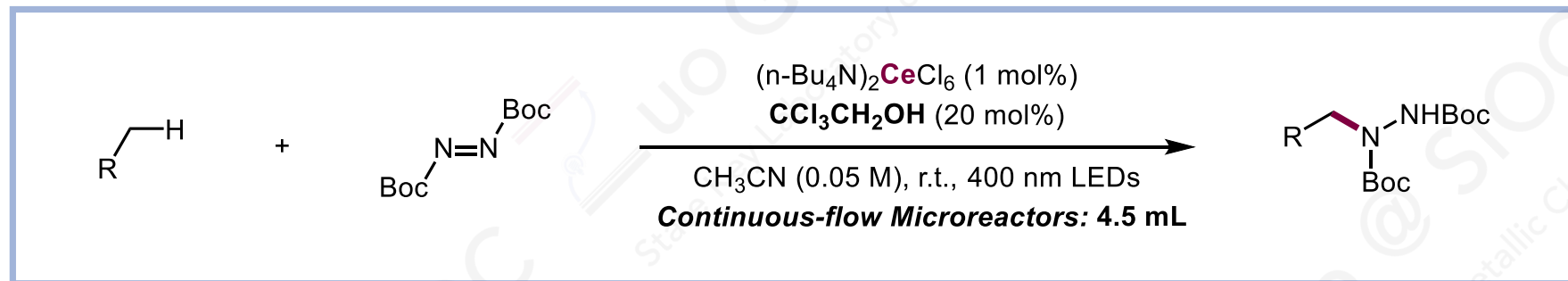


| Entry <sup>a</sup> | $P_{\text{CH}_4}$ | $V_{\text{G}}$<br>(mL/min) | $V_{\text{L}}$<br>(mL/min) | $t_{\text{R}}$<br>(min) | Yield<br>(%) | Productivity<br>(mmol/d) |
|--------------------|-------------------|----------------------------|----------------------------|-------------------------|--------------|--------------------------|
| 1                  | 13 bar            | 10.0                       | 0.75                       | 6                       | 9            | 1.0                      |
| 2                  |                   | 2.0                        | 0.30                       | 15                      | 12           | 0.52                     |
| 3                  |                   | 1.5                        | 0.15                       | 30                      | 14           | 0.30                     |
| 4                  | 18 bar            | 1.5                        | 0.15                       | 30                      | 15           | 0.32                     |
| 5                  |                   | 6.0                        | 0.30                       | 15                      | 15           | 0.60                     |



$$t_{\text{R}} = \frac{V}{V_{\text{G}}/P + V_{\text{L}}}$$

## 3.2 铈-光催化的轻质烷烃胺化反应

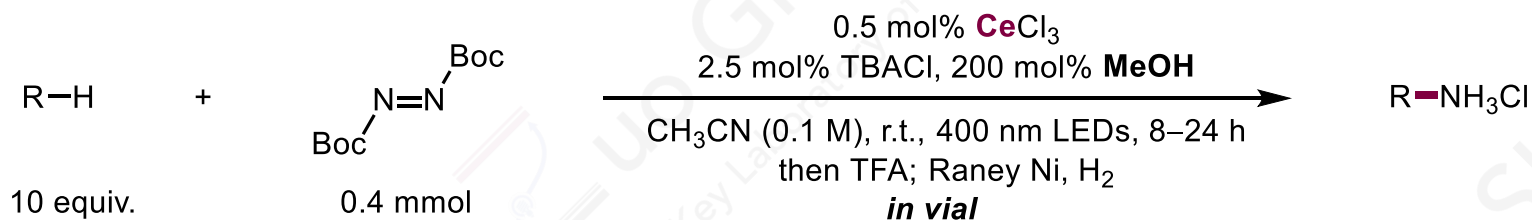


| Entry          | Alkane              | $V_G$<br>(mL/min) | $V_L$<br>(mL/min) | $t_R$<br>(min) | Yield<br>(%) | Production<br>(mmol/d) | Production in<br>batch (mmol/d) |
|----------------|---------------------|-------------------|-------------------|----------------|--------------|------------------------|---------------------------------|
| 1 <sup>a</sup> | Ethane (15 bar)     | 24                | 0.75              | 6              | 90           | 48                     | 0.59                            |
| 2 <sup>b</sup> | Propane (8 bar)     | 24                | 0.75              | 6              | 76           | 41                     | 0.74                            |
| 3 <sup>c</sup> | Butane (4 bar)      | 30                | 0.75              | 6              | 56           | 31                     | 1.2                             |
| 4 <sup>d</sup> | cyclohexane (22 mL) | —                 | 0.5               | 9              | 70           | 100                    | 0.48                            |

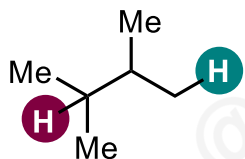
<sup>a</sup> DIAD instead of DBAD. <sup>b</sup> 2° : 1° = 1 : 1. <sup>c</sup> 2° : 1° = 1.8 : 1. <sup>d</sup> In liquid mode at 0.2 M.

Z. Zuo, et al. *Science* **2018**, 361, 668–672.

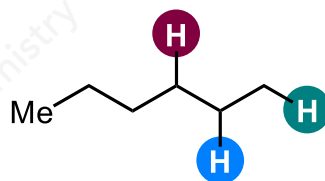
### 3.3 铈-光催化的烷烃选择性胺化反应



#### Substrate scope



MeOH: 81%, **97** : **3**  
*t*BuOH: 49%, **64** : **36**  
 CCl<sub>3</sub>CH<sub>2</sub>OH: 83%, **44** : **56**



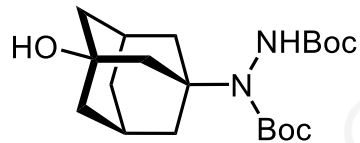
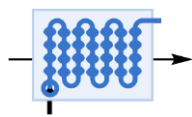
MeOH: 82%, **47** : **48** : **3**  
 CCl<sub>3</sub>CH<sub>2</sub>OH: 85%, **39** : **39** : **22**



72% yield

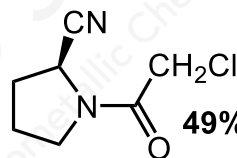


79% yield, 9 h

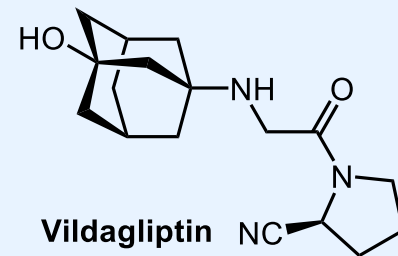


2 g scale

2) TFA; Raney Ni, H<sub>2</sub>  
 3) K<sub>2</sub>CO<sub>3</sub>, TBAI, 60°C

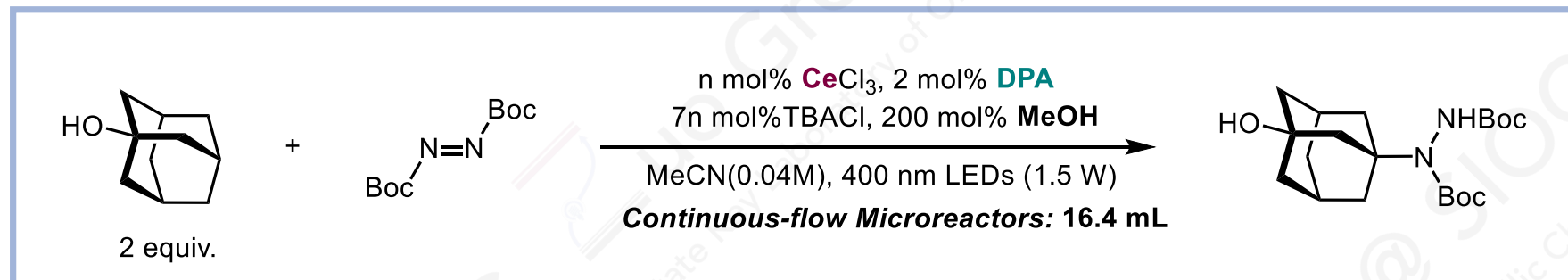


49% for 3 steps



**Vildagliptin**  
 antidiabetic agent

### 3.3 铈-光催化的烷烃选择性胺化反应

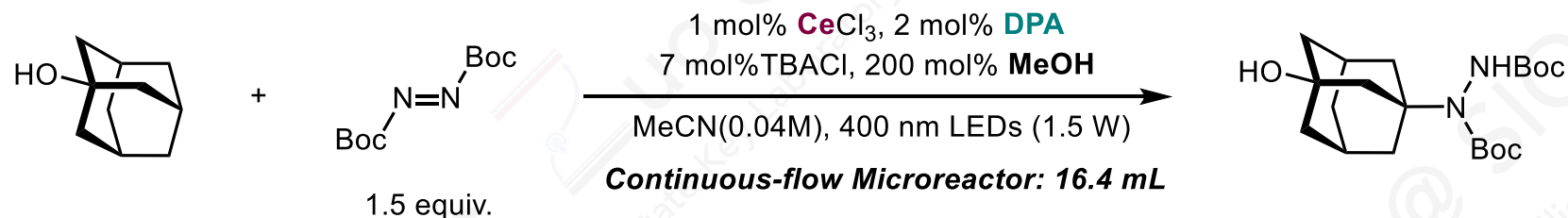


| Entry                | $V_L$<br>(mL/min) | $t_R$<br>(min) | DPA           | $\text{CeCl}_3$           | MeOH            | Yield<br>(%) | Productivity<br>(mmol/d) |
|----------------------|-------------------|----------------|---------------|---------------------------|-----------------|--------------|--------------------------|
| 1                    | 2.0               | 8.2            | —             | $n = 1$                   | 200 mol%        | 14%          | 16                       |
| 2                    |                   |                | <b>2 mol%</b> | <b><math>n = 1</math></b> | <b>200 mol%</b> | <b>40%</b>   | <b>46</b>                |
| 3                    |                   |                | 2 mol%        | $n = 1$                   | 400 mol%        | 42%          | 48                       |
| 4                    |                   |                | 2 mol%        | $n = 5$                   | 200 mol%        | 44%          | 50                       |
| <b>5<sup>a</sup></b> | 1.0               | 16             | <b>2 mol%</b> | $n = 1$                   | <b>200 mol%</b> | <b>56%</b>   | <b>34</b>                |
| 6 <sup>b</sup>       |                   |                | 2 mol%        | $n = 1$                   | 200 mol%        | 57%          | 34                       |
| 7                    |                   |                | 5 mol%        | $n = 1$                   | 200 mol%        | 58%          | 34                       |

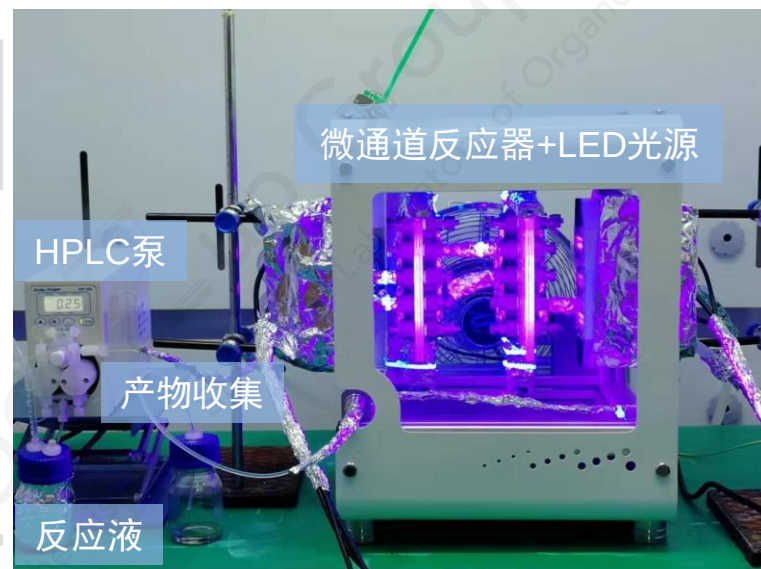
<sup>a</sup> 4% DBAD remained. <sup>b</sup> 400 nm LEDs (2.0 W )

Z. Zuo, et al. *J. Am. Chem. Soc.* **2020**, 142, 6216–6226.

### 3.3 铈-光催化的烷烃选择性胺化反应



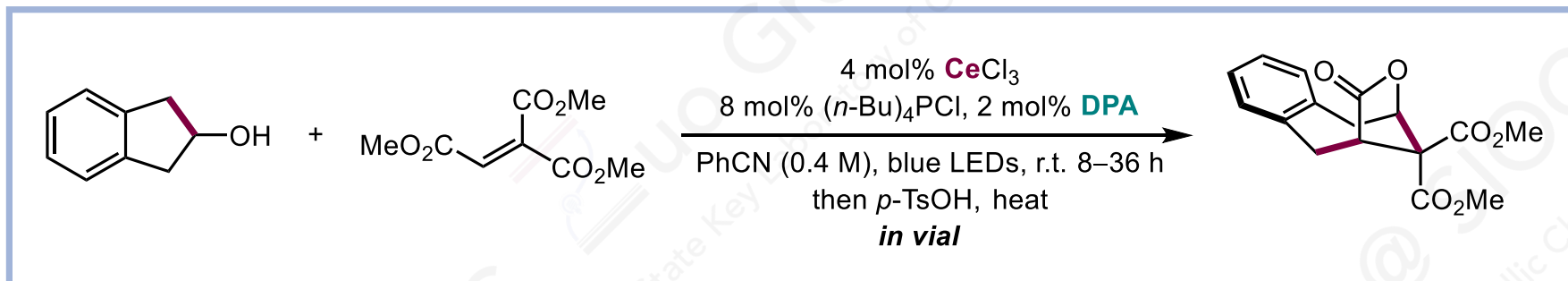
| Entry    | V <sub>L</sub><br>(mL/min) | t <sub>R</sub><br>(min) | DBAD<br>(%) | Yield<br>(%) | Productivity<br>(mmol/d) |
|----------|----------------------------|-------------------------|-------------|--------------|--------------------------|
| 1        | 1.0                        | 16                      | 38%         | <b>50%</b>   | <b>29</b>                |
| 2        | 0.5                        | 33                      | 20%         | <b>59%</b>   | <b>17</b>                |
| <b>3</b> | <b>0.25</b>                | <b>66</b>               | <b>2%</b>   | <b>66%</b>   | <b>9.4</b>               |



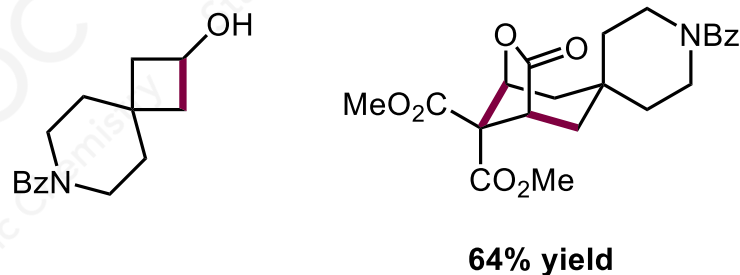
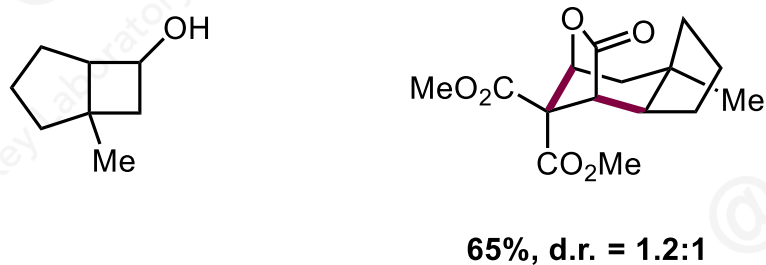
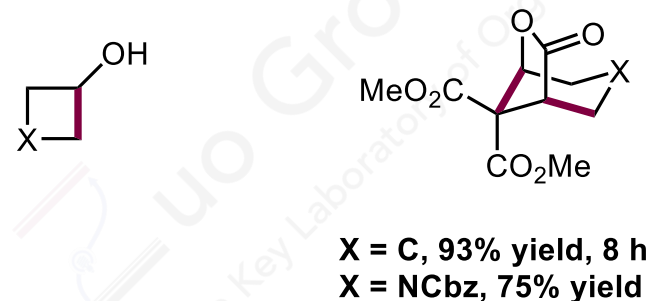
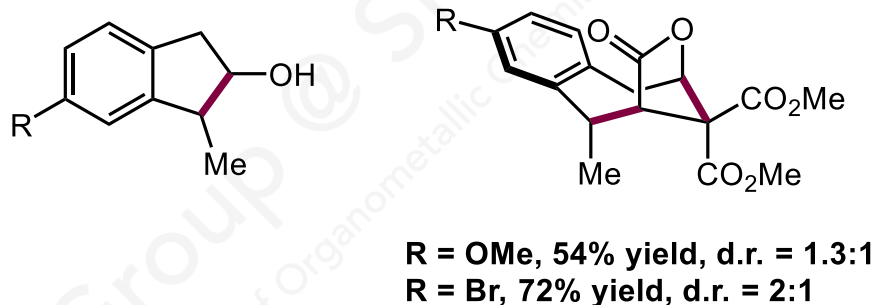
Z. Zuo, et al. *J. Am. Chem. Soc.* **2020**, 142, 6216–6226.



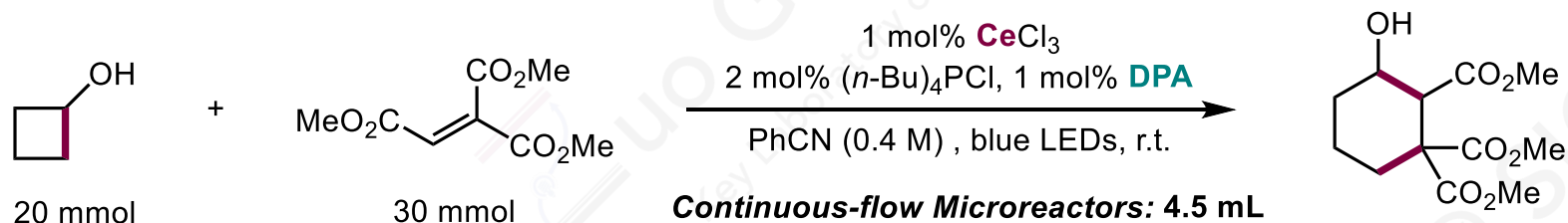
### 3.4 铈-光催化的环醇和烯烃形式环加成反应



#### Substrate scope



### 3.4 铈-光催化的环醇和烯烃形式环加成反应

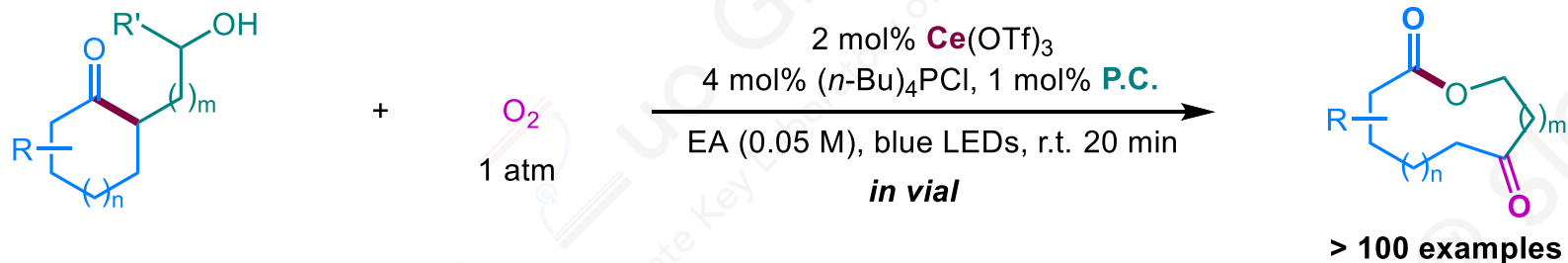


| Entry          | Mode         | $V_L$<br>(mL/min) | $t_R$ (h) | Yield<br>(%) | Productivity<br>(mmol/d)  |
|----------------|--------------|-------------------|-----------|--------------|---------------------------|
| 1              | single-pass  | 0.15              | 0.5       | 11%          | 9.6                       |
| 2 <sup>a</sup> | recirculated | 1.5               | —         | 80%          | <b>10.6</b> (batch: 0.57) |

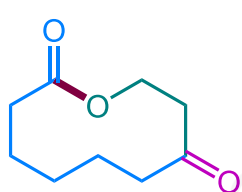
<sup>a</sup> Reaction time: 36 h.



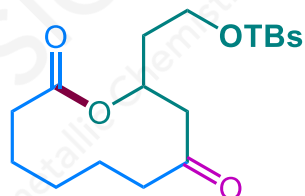
### 3.5 铈-光催化的环酮氧化扩环反应



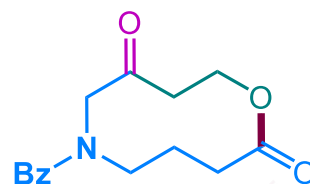
#### Substrate scope



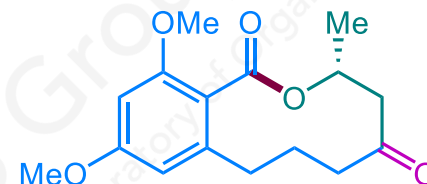
$n = 1$ , 60% yield  
 $n = 5$ , 60% yield  
 $n = 9$ , 72% yield



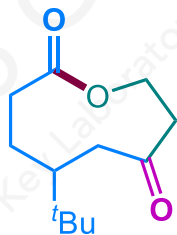
74% yield



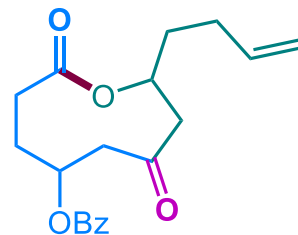
68% yield



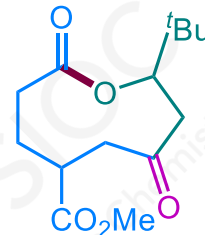
77% yield



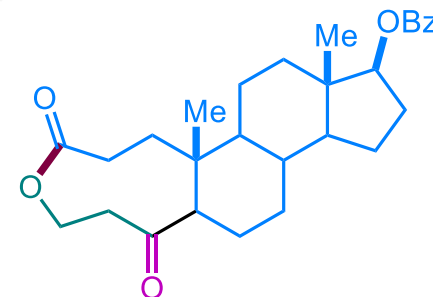
66% yield



54% yield, 1.5:1 d.r.

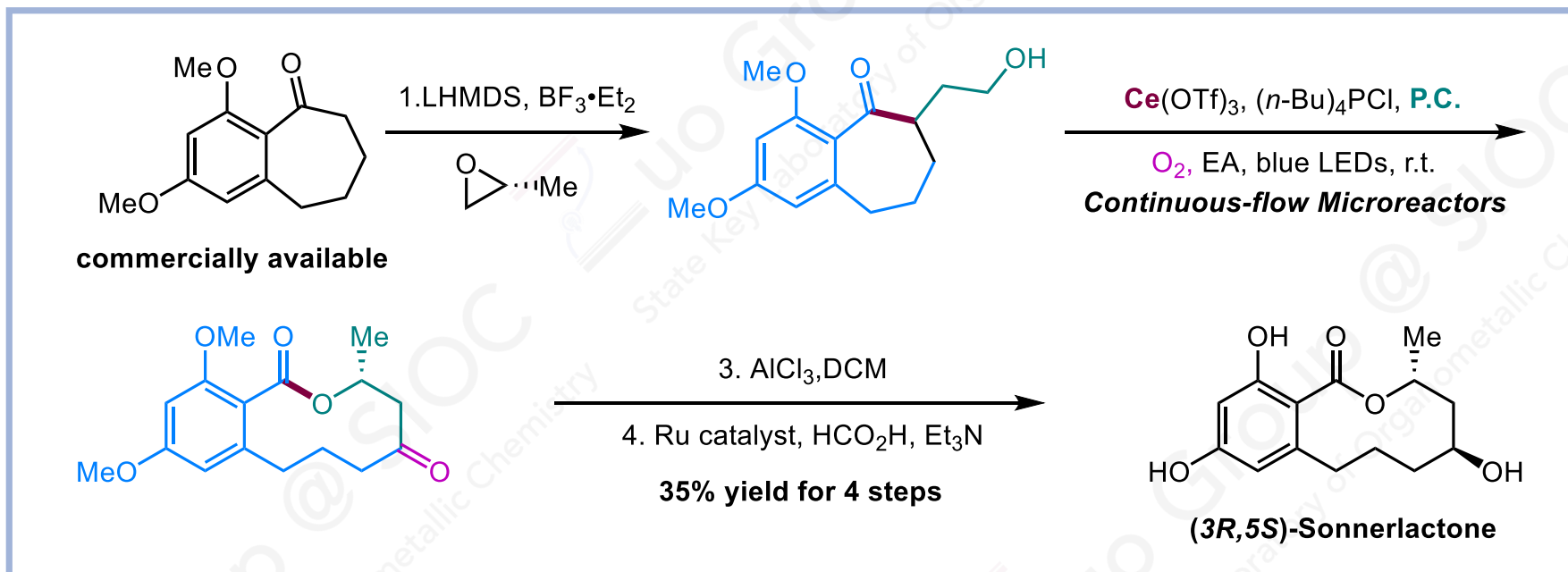


65% yield, 1.8:1 d.r.

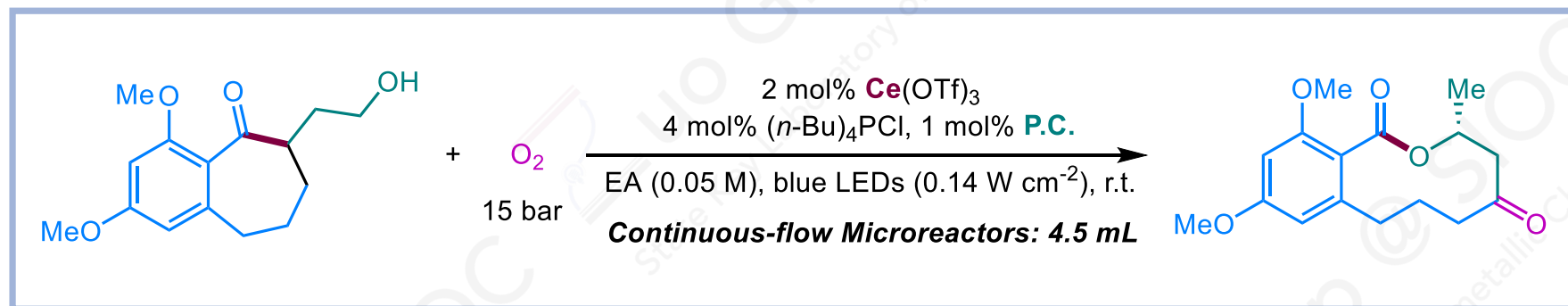


73% yield

### 3.5 铈-光催化的环酮扩环反应



### 3.5 铈-光催化的环酮扩环反应



| Entry <sup>a</sup>   | $V_L$<br>(mL/min) | $V_g$<br>(mL/min) | $t_R$<br>(min) | Yield<br>(%) | Productivity<br>(mmol/d) |
|----------------------|-------------------|-------------------|----------------|--------------|--------------------------|
| 1                    | 0.40              | 1.0               | 10             | 55%          | 17.8                     |
| 2                    | 0.30              | 1.0               | 12             | 65%          | 17.5                     |
| <b>3<sup>a</sup></b> | <b>0.15</b>       | <b>1.0</b>        | <b>20</b>      | <b>78%</b>   | <b>15.3</b> (batch: 5.5) |

<sup>a</sup> isolated yield.

Z. Zuo, et al. *Angew. Chem. Int. Ed.* **2021**, 60, 5370–5376.